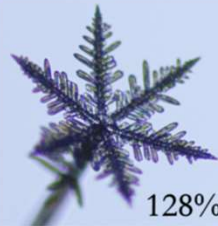


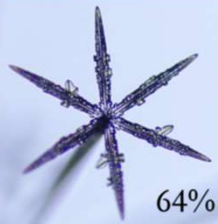
Snow Crystals: I

A Case Study in Spontaneous Structure Formation

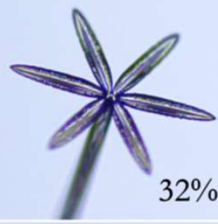
**Kenneth G. Libbrecht
Dept. of Physics
Caltech**



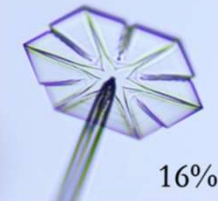
128%



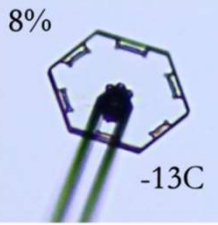
64%



32%



16%



8%
-13C



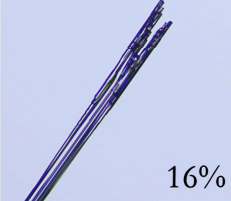
128%



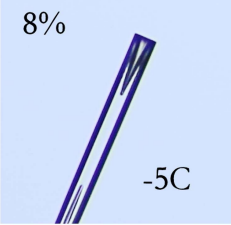
64%



32%



16%

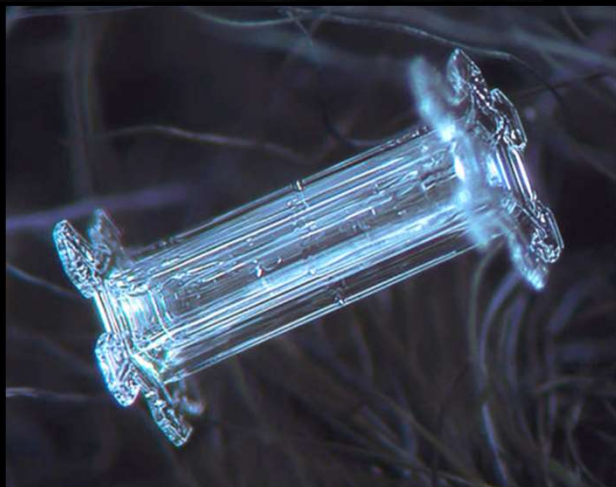
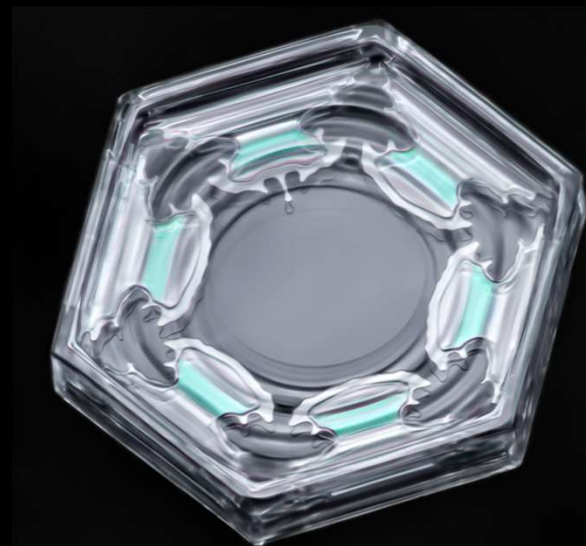


8%

-5C



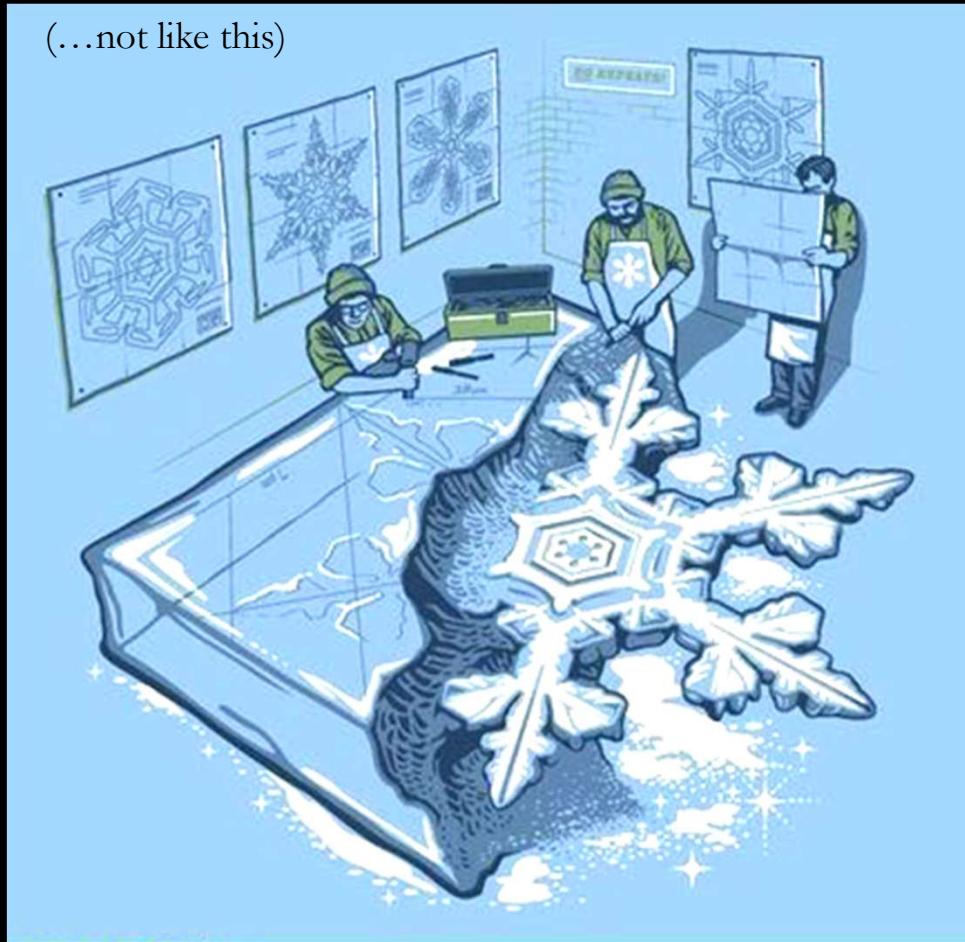
Don Komarechka
Don Komarechka



Nature makes things using self-assembly

(...not like this)

An
archetypal
example of
large-scale
structural
self-assembly



Christopher Buchholz



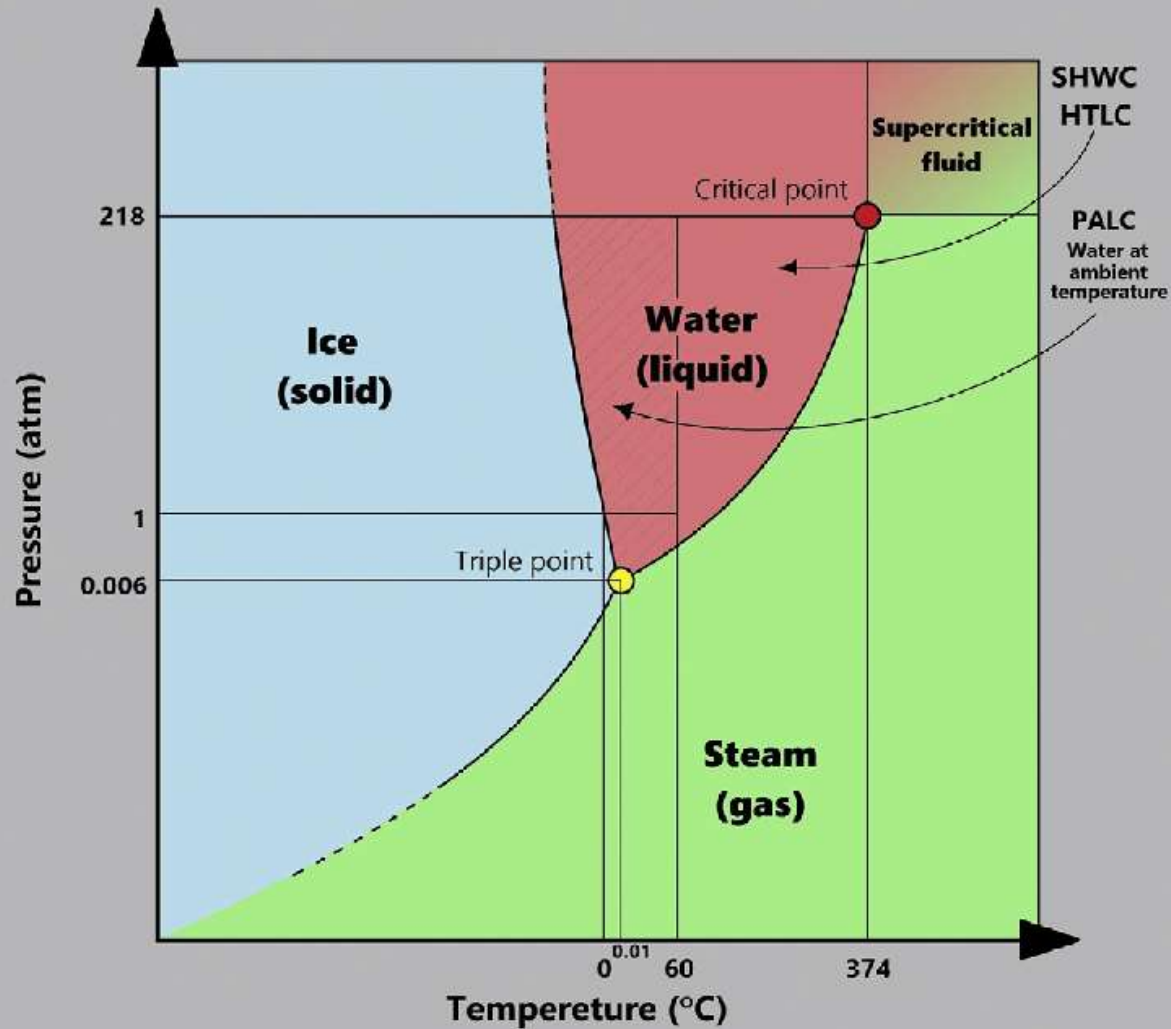
Can we “fully” understand
even a simple system?

= Make a computational model
that reasonably reproduces
known growth behaviors,
agrees with experiments.
→ Model relevant physical processes

Ice/vapor: An interesting phase transition

Ice/vapor
(T, σ)

Ice/water
($P, \Delta T$)
 $\rightarrow (P_0, \Delta T)$



A laboratory “Snowflake on a Stick”



Environment:

Fixed temperature – $T < 0\text{ C}$

Fixed supersaturation – $\sigma > 0$
(RH > 100%)

In air at 1 atm

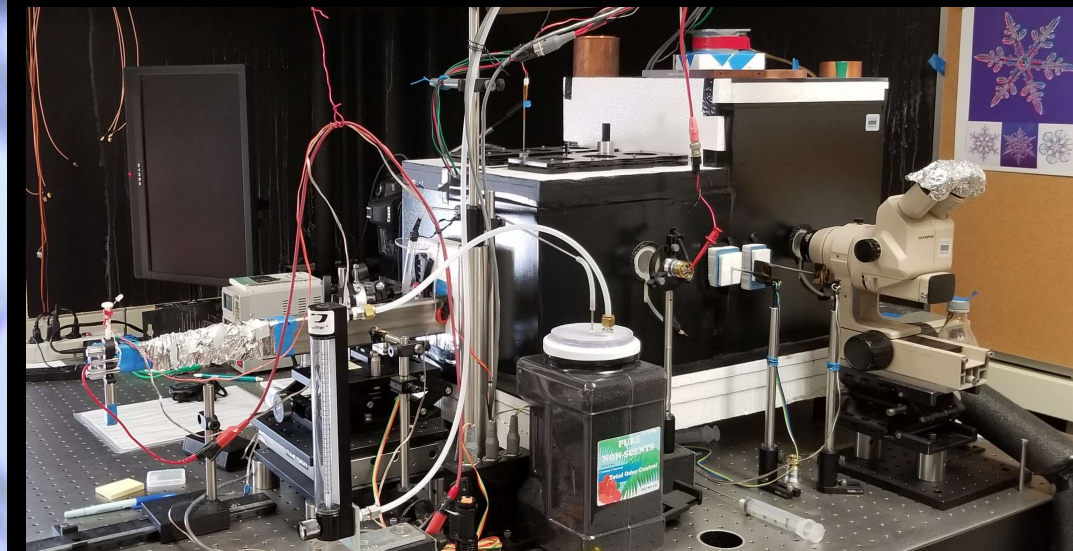
Add seed crystal

Thin c-axis ice needle

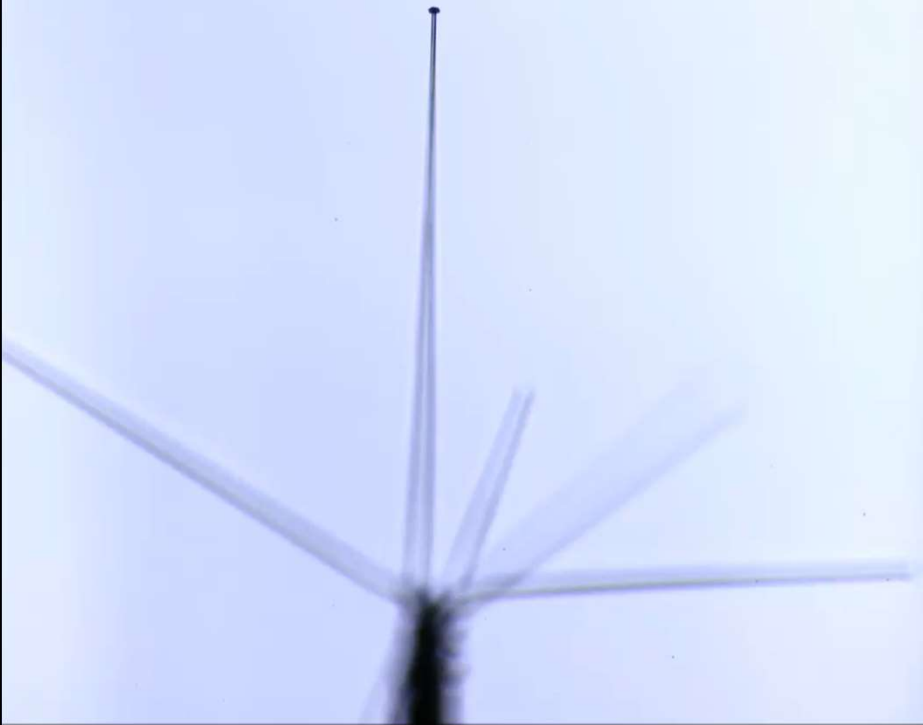
~ 2 mm long, ~2 μm tip

This example:

$T = -15\text{ C}$, $\sigma = 16\%$
then 64%



A laboratory “Snowflake on a Stick”



Environment:

Fixed temperature – $T < 0\text{ }^{\circ}\text{C}$

Fixed supersaturation – $\sigma > 0$
(RH > 100%)

In air at 1 atm

Add seed crystal

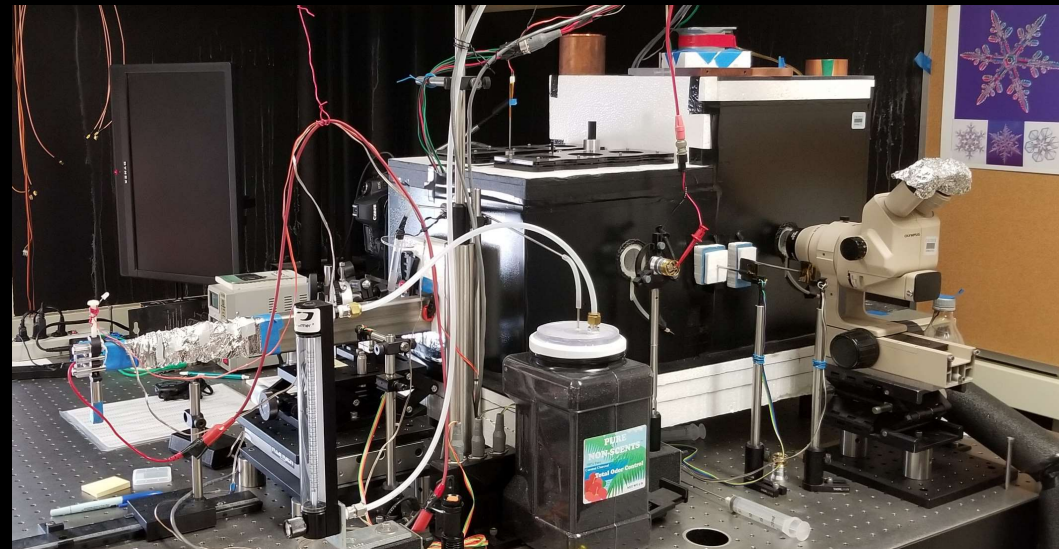
Thin ice needle

~ 2 mm long, ~2 μm tip

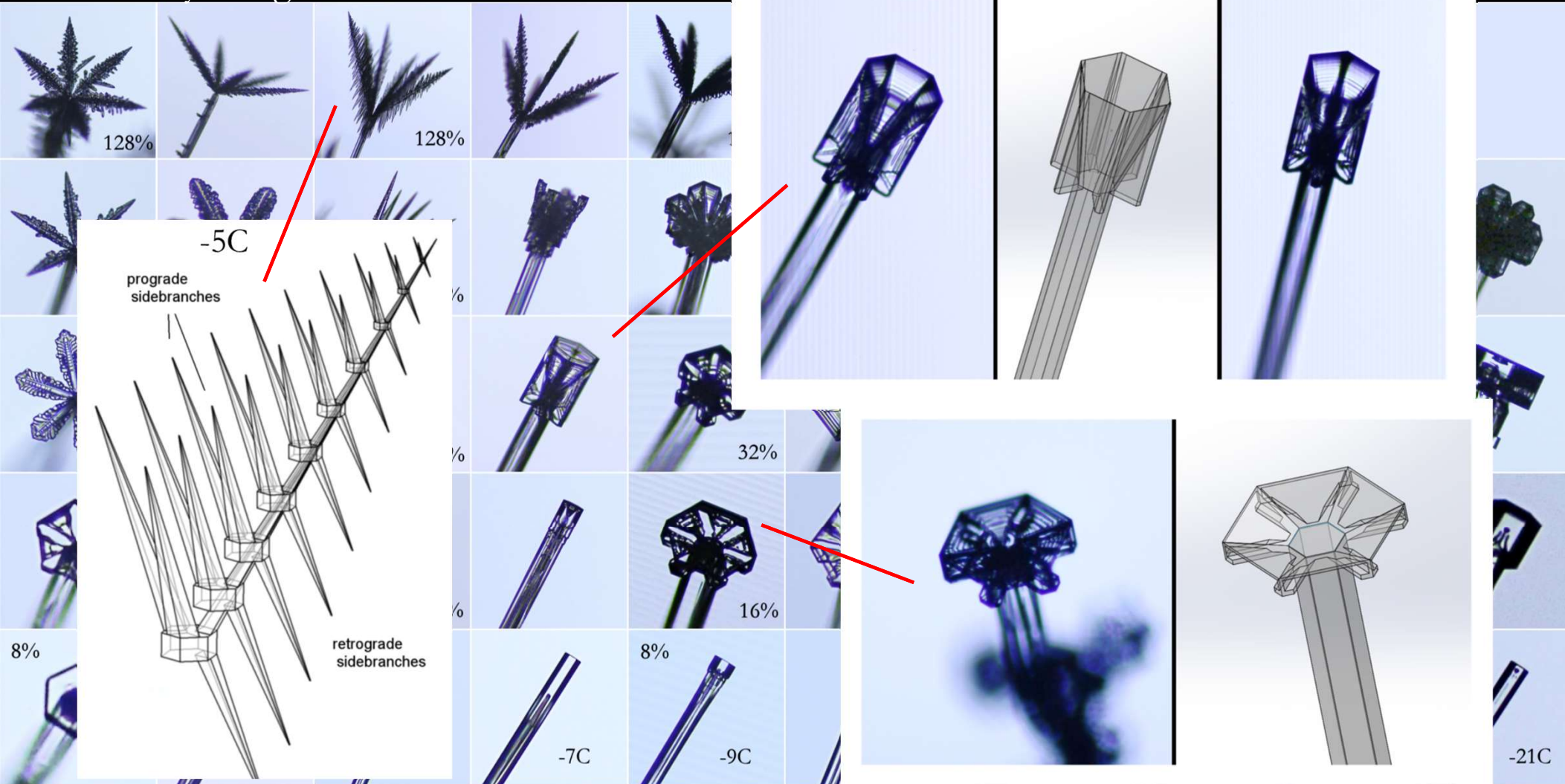
This example:

$T = -15\text{ }^{\circ}\text{C}$, $\sigma = 16\%$
then 64%

Growth time ~20 minutes



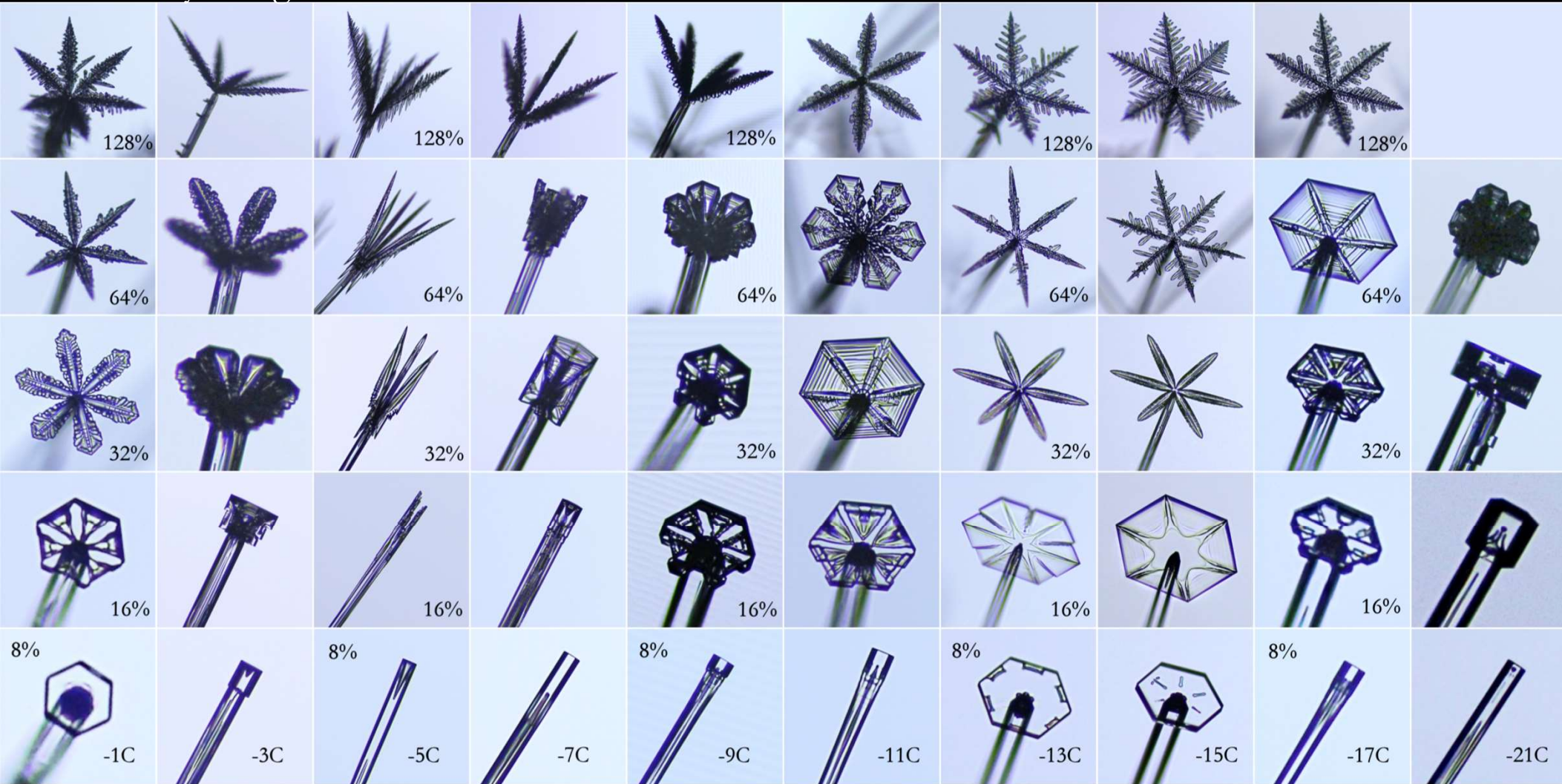
Snow crystals grown on ice needles – at different



All single crystals; all grown at constant conditions

A complex physical phenomenon → complex physics

Snow crystals grown on ice needles



All single crystals; all grown at constant conditions.

Possible to make numerical models? What is underlying physics?

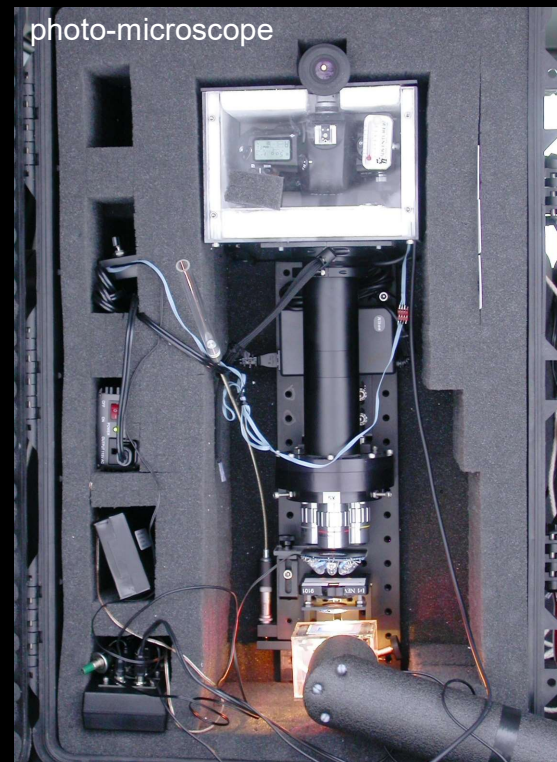
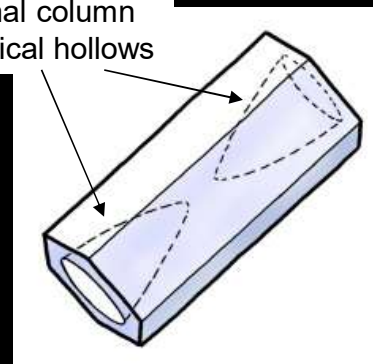
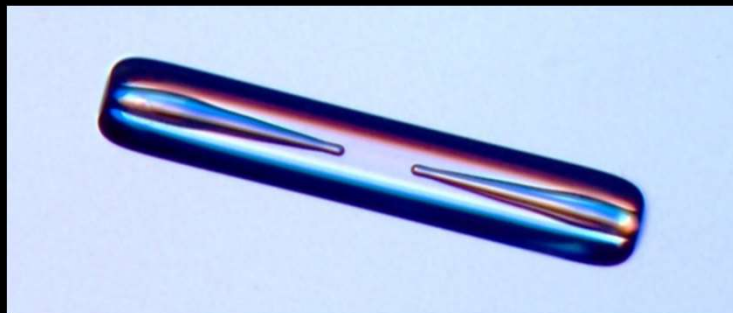
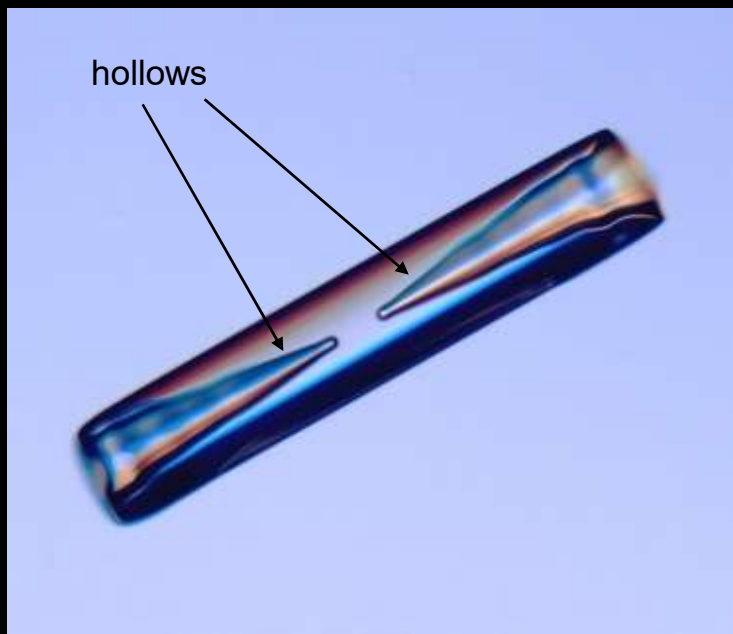
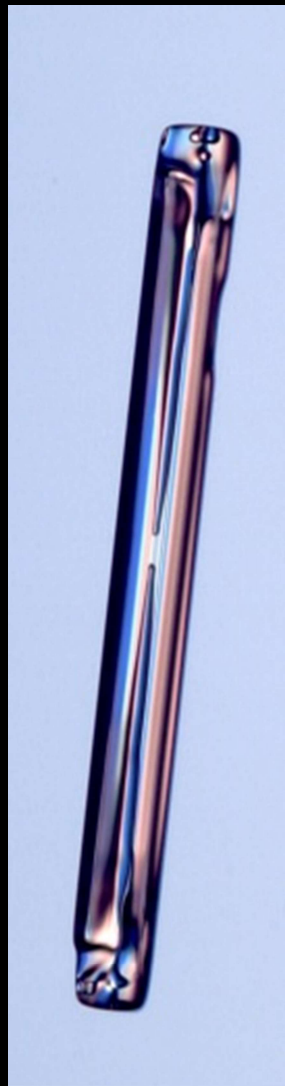
Natural Snow Crystals...

Hollow Columns, form at -5°C

hollows

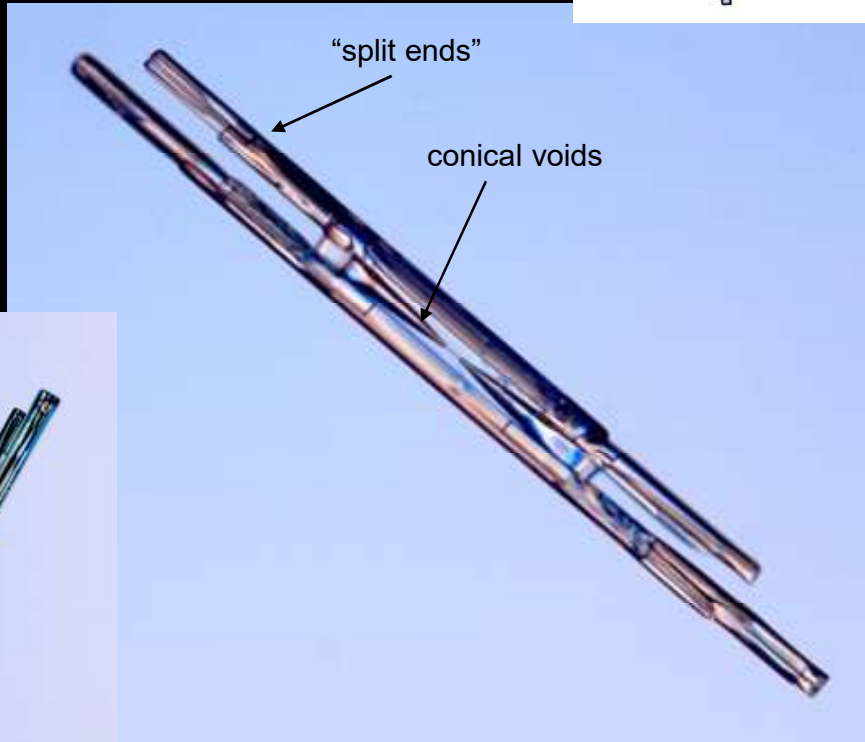
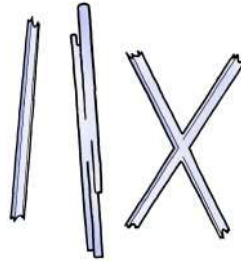
hexagonal column
with conical hollows

photo-microscope

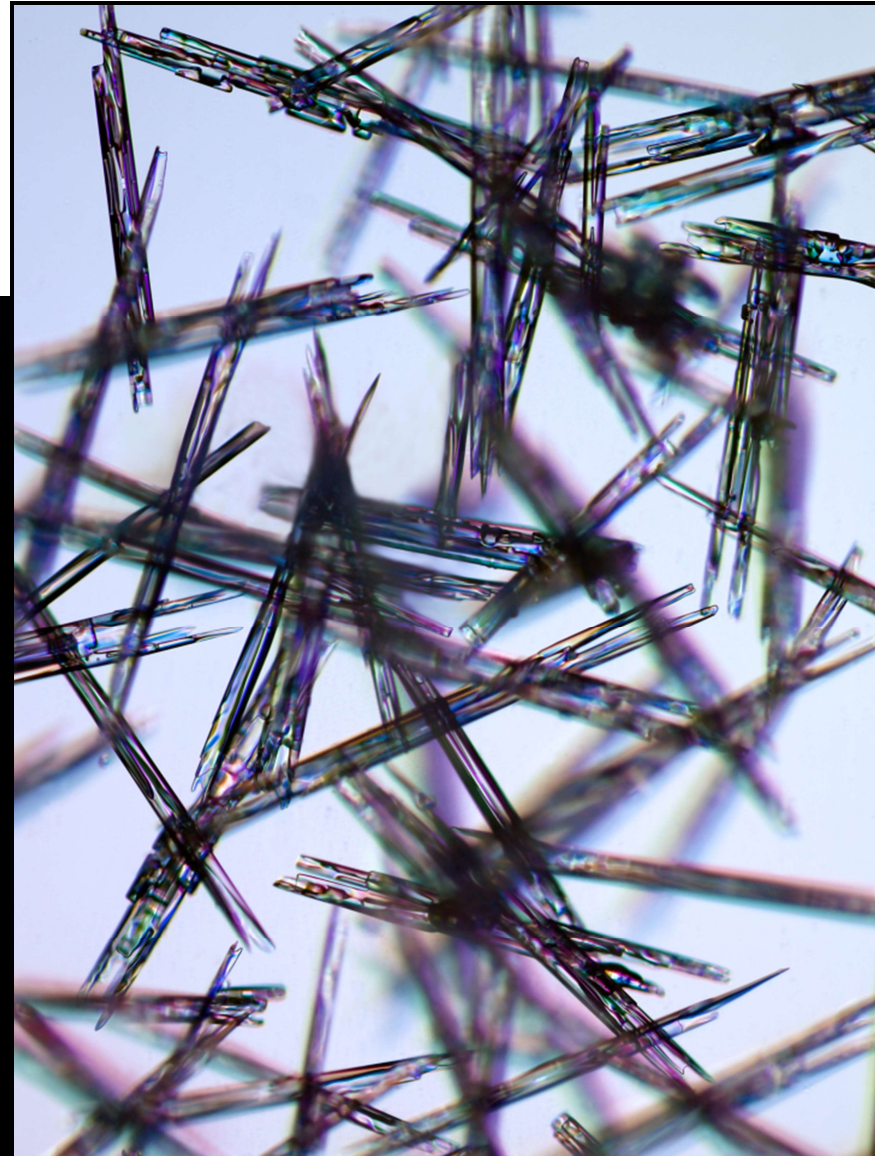


Natural Snow Crystals...

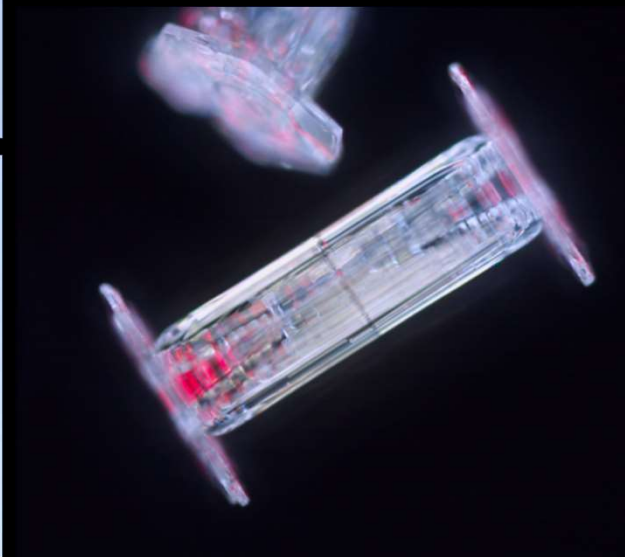
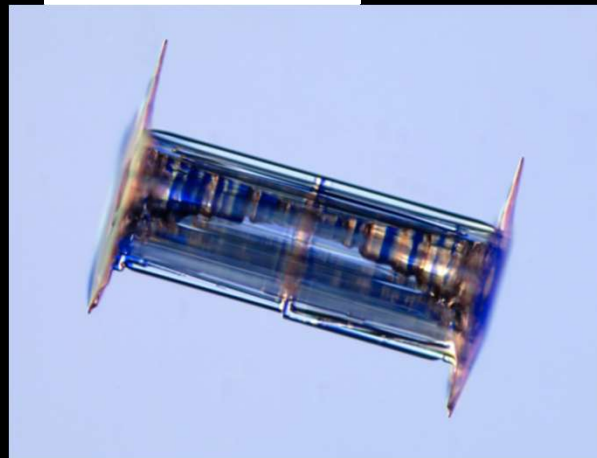
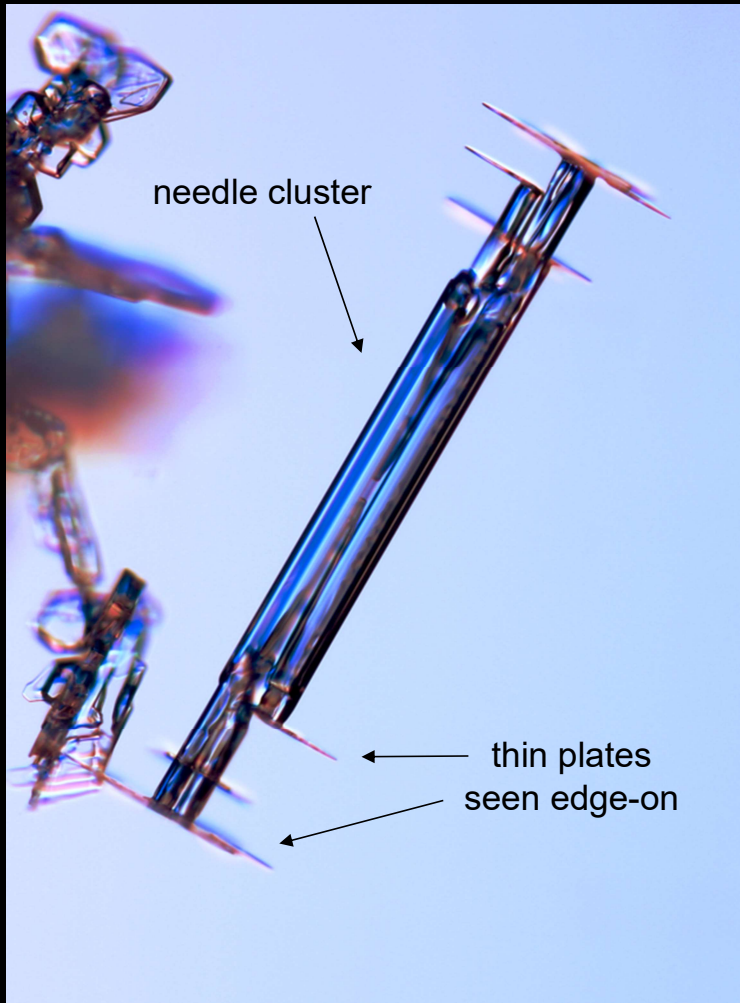
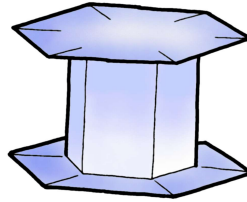
Needle Crystals, form at -5°C



longest columnar crystals 2-3 mm
often find in clusters

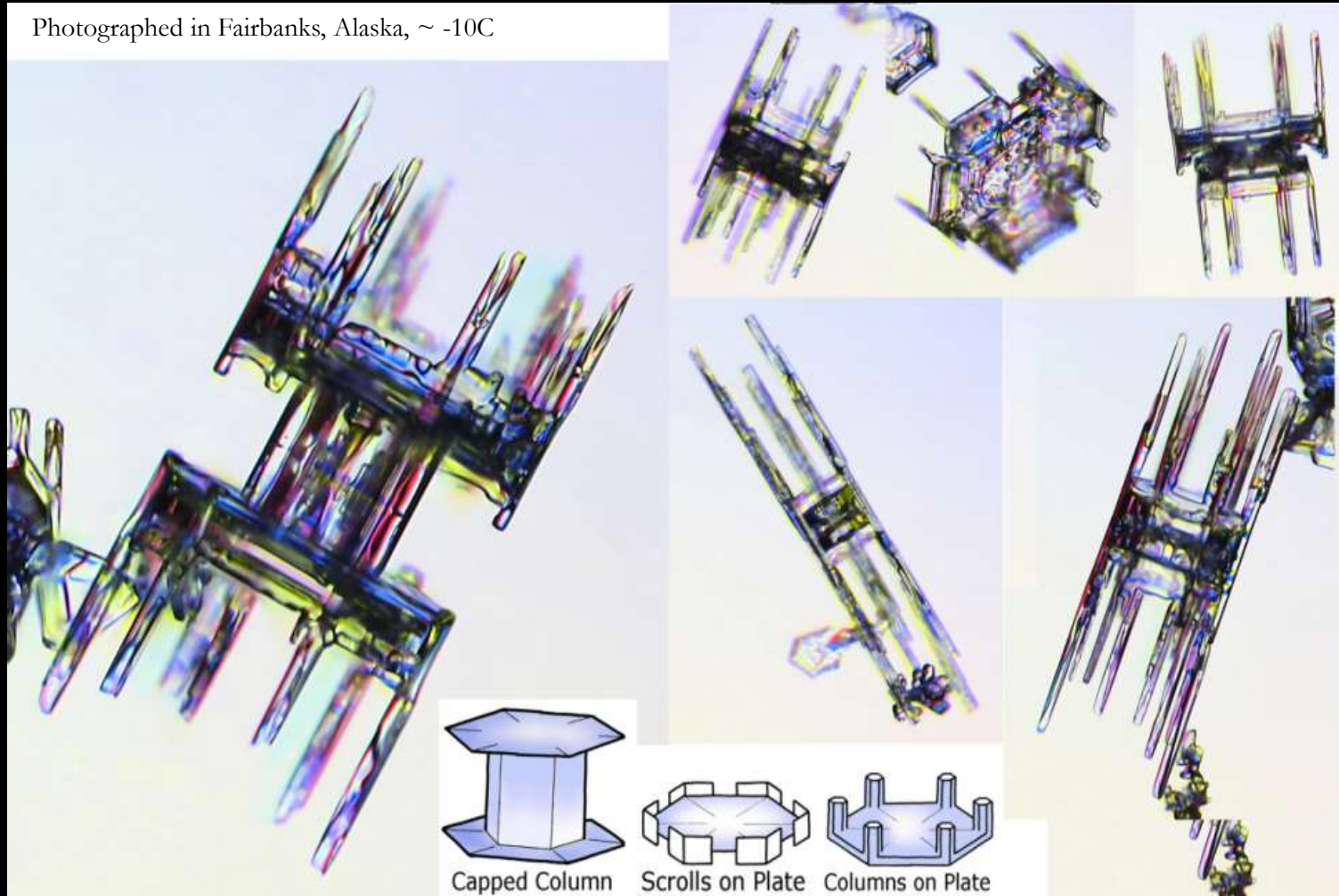


Natural Snow Crystals...
Capped Columns, -6C → -13C



Some especially odd examples: Capped columns with scrolls and columns

Photographed in Fairbanks, Alaska, ~ -10C

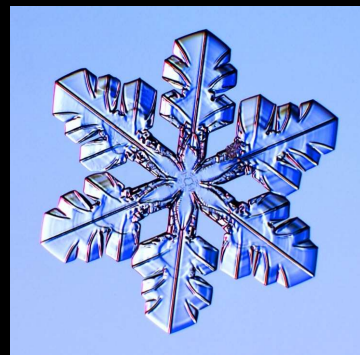
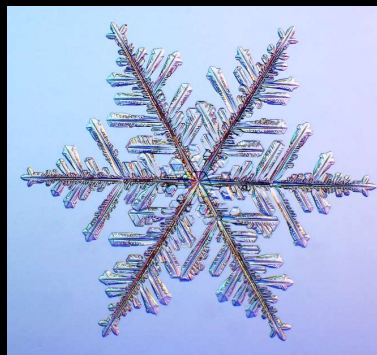
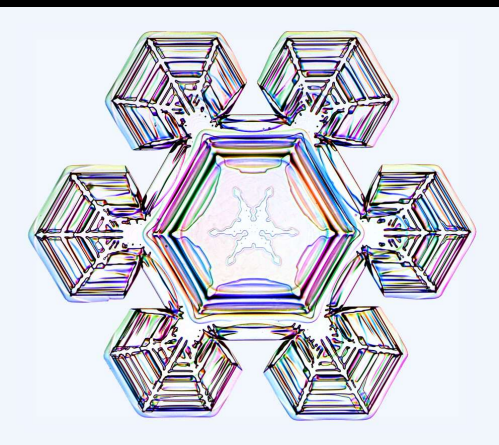
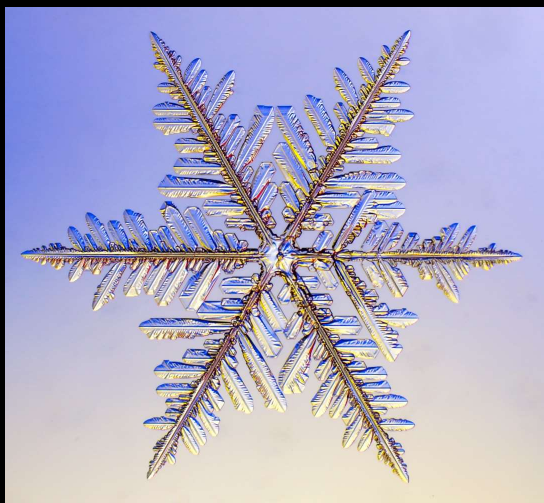


Where to start?....

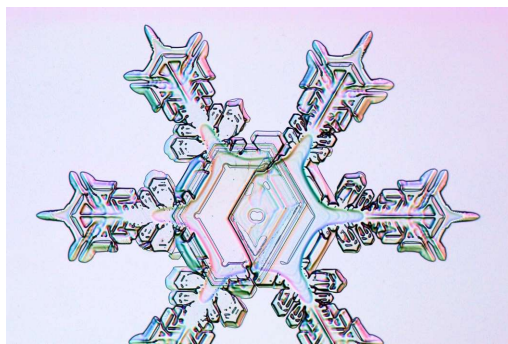
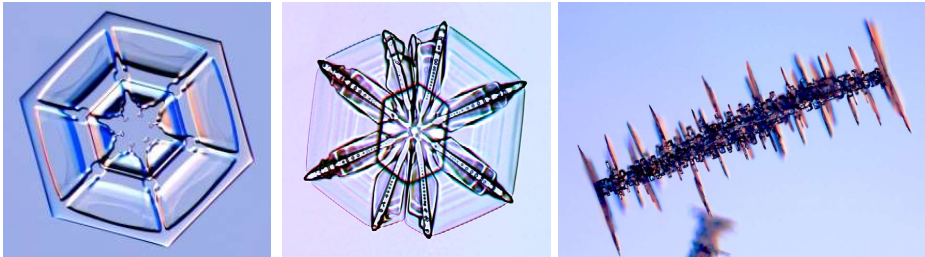
Natural Snow Crystals...

Stellar Crystals, form at -15°C




(branched, faceted, thin and flat)



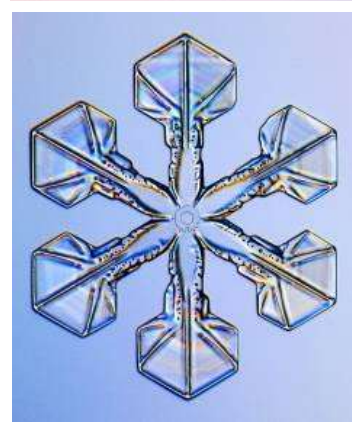
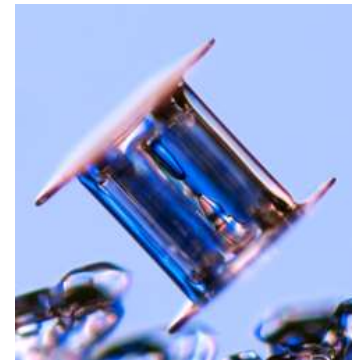
A Menagerie of Natural Snow Crystals...

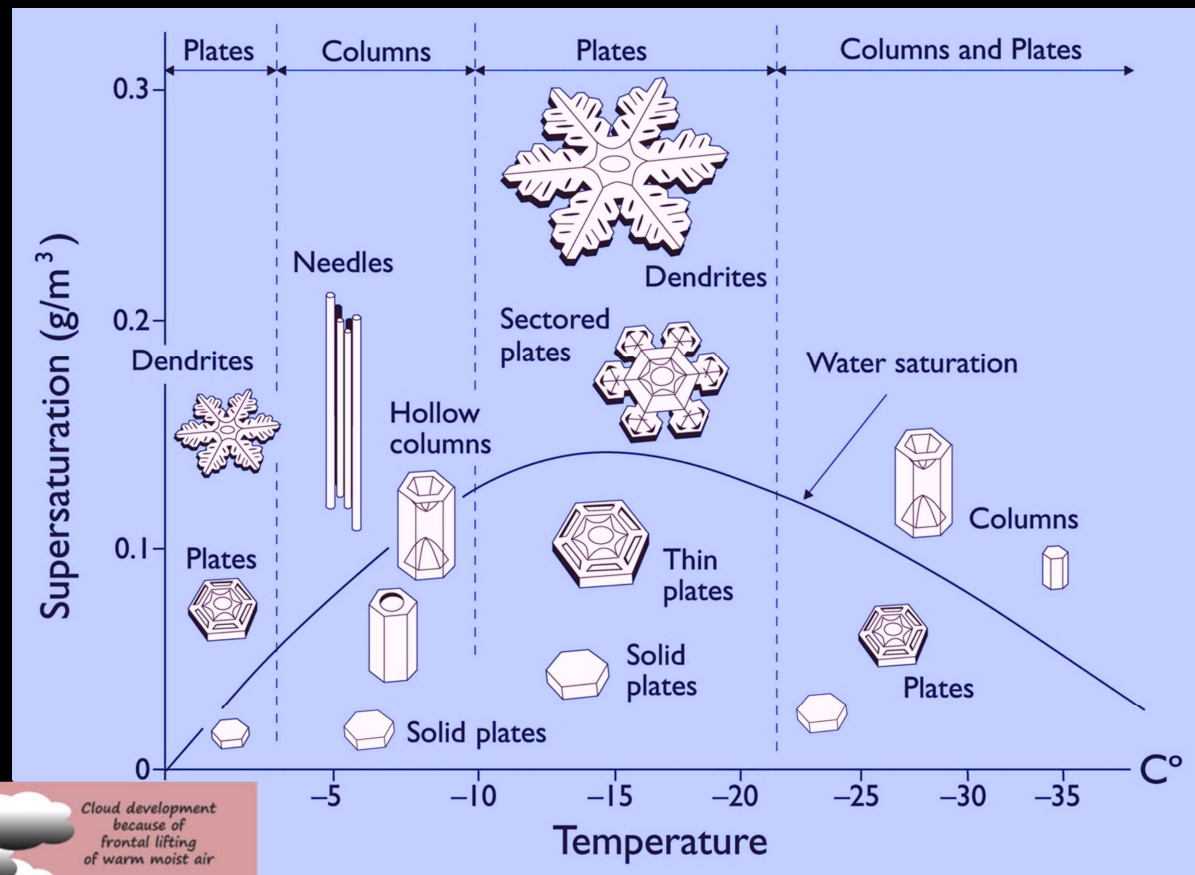
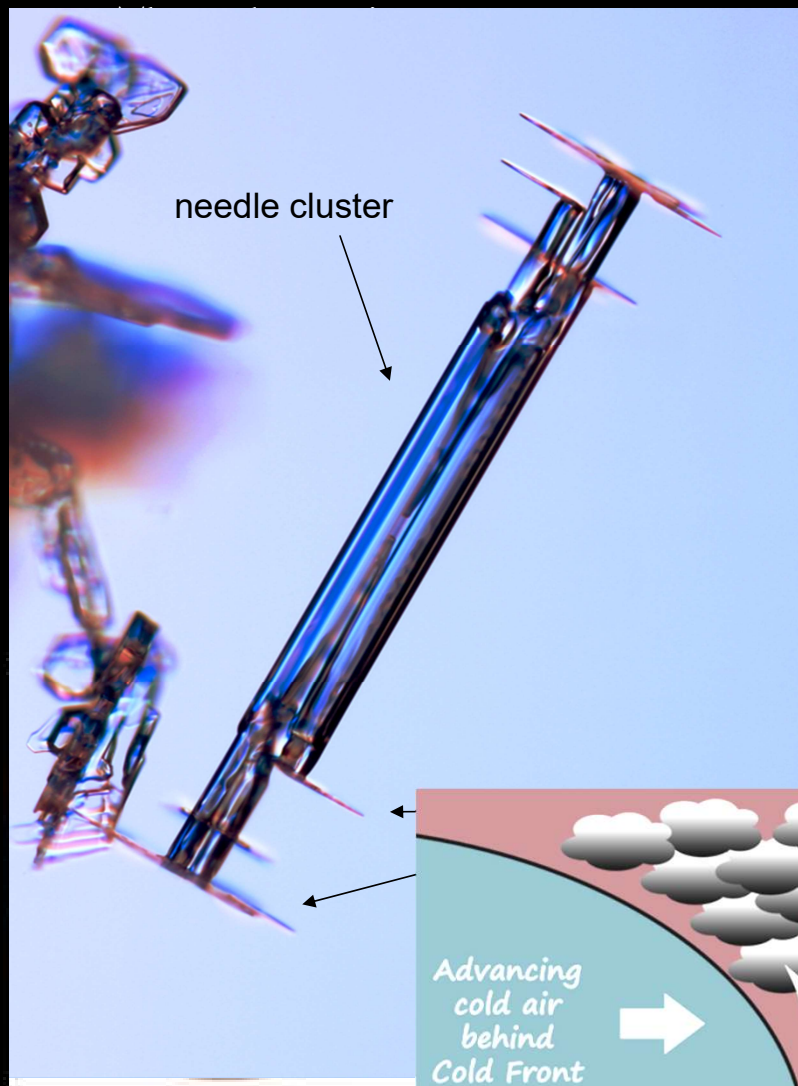


What is the underlying physics?

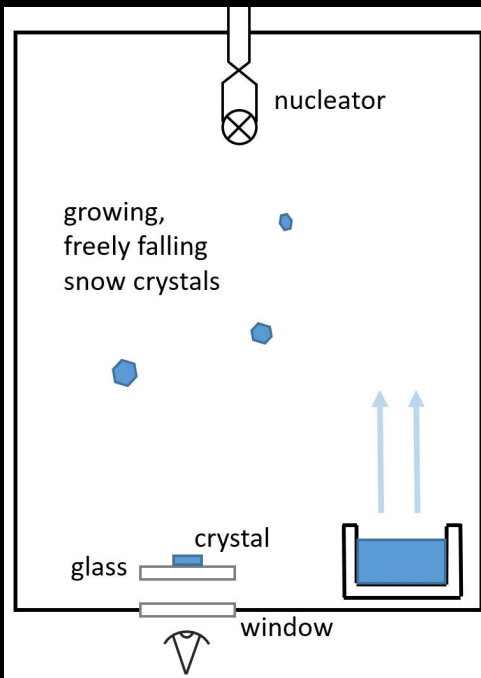
				
Simple Prisms	Solid Columns	Sheaths	Scrolls on Plates	Triangular Forms
				
Hexagonal Plates	Hollow Columns	Cups	Columns on Plates	12-branched Stars
				
Stellar Plates	Bullet Rosettes	Capped Columns	Split Plates & Stars	Radiating Plates
				
Sectorial Plates	Isolated Bullets	Multiply Capped Columns	Skeletal Forms	Radiating Dendrites
				
Simple Stars	Simple Needles	Capped Bullets	Twin Columns	Irregulars
				
Stellar Dendrites	Needle Clusters	Double Plates	Arrowhead Twins	Rimed
				
Fernlike Stellar Dendrites	Crossed Needles	Hollow Plates	Crossed Plates	Graupel

from *Ken Libbrecht's Field Guide to Snowflakes*

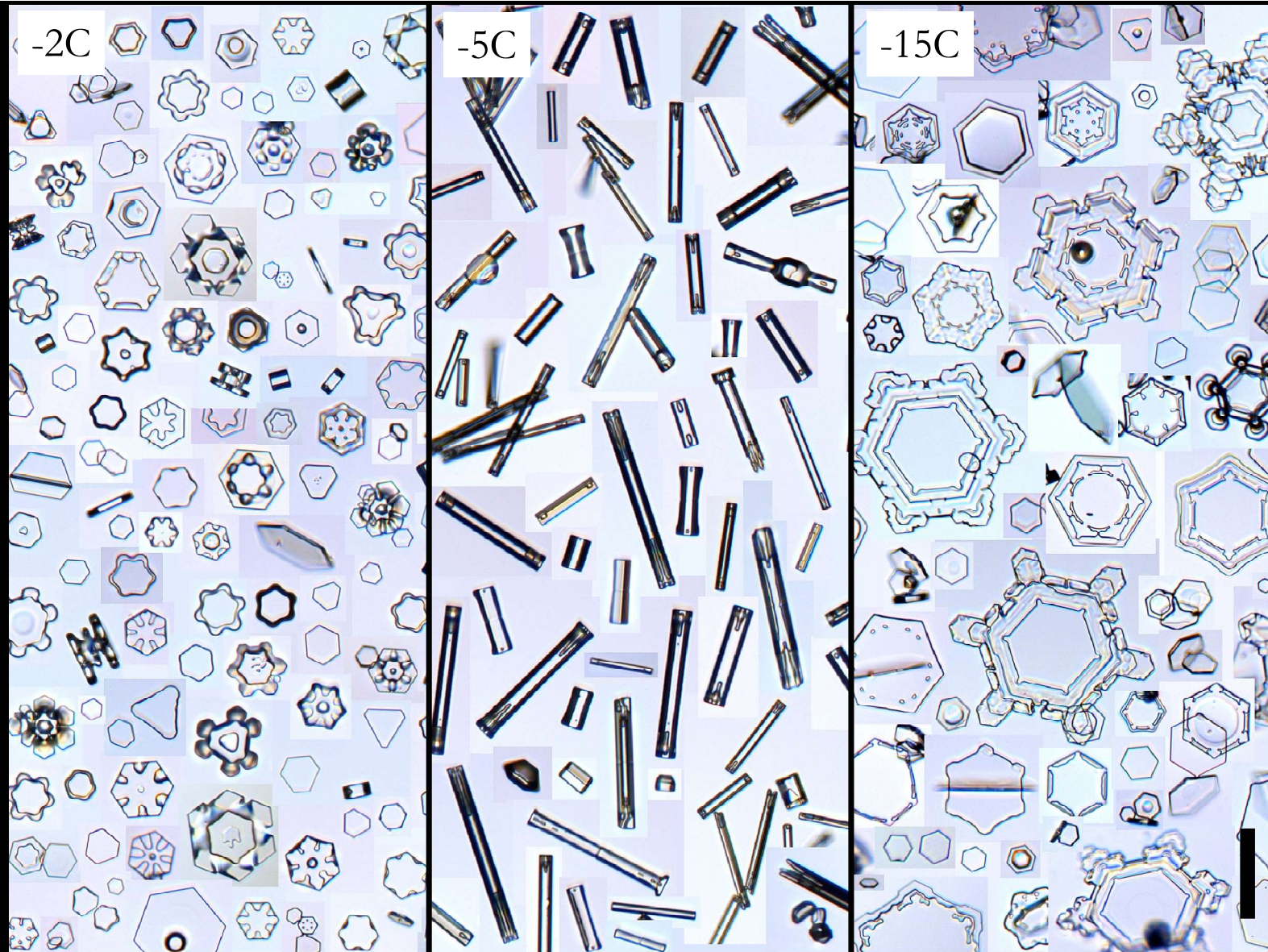




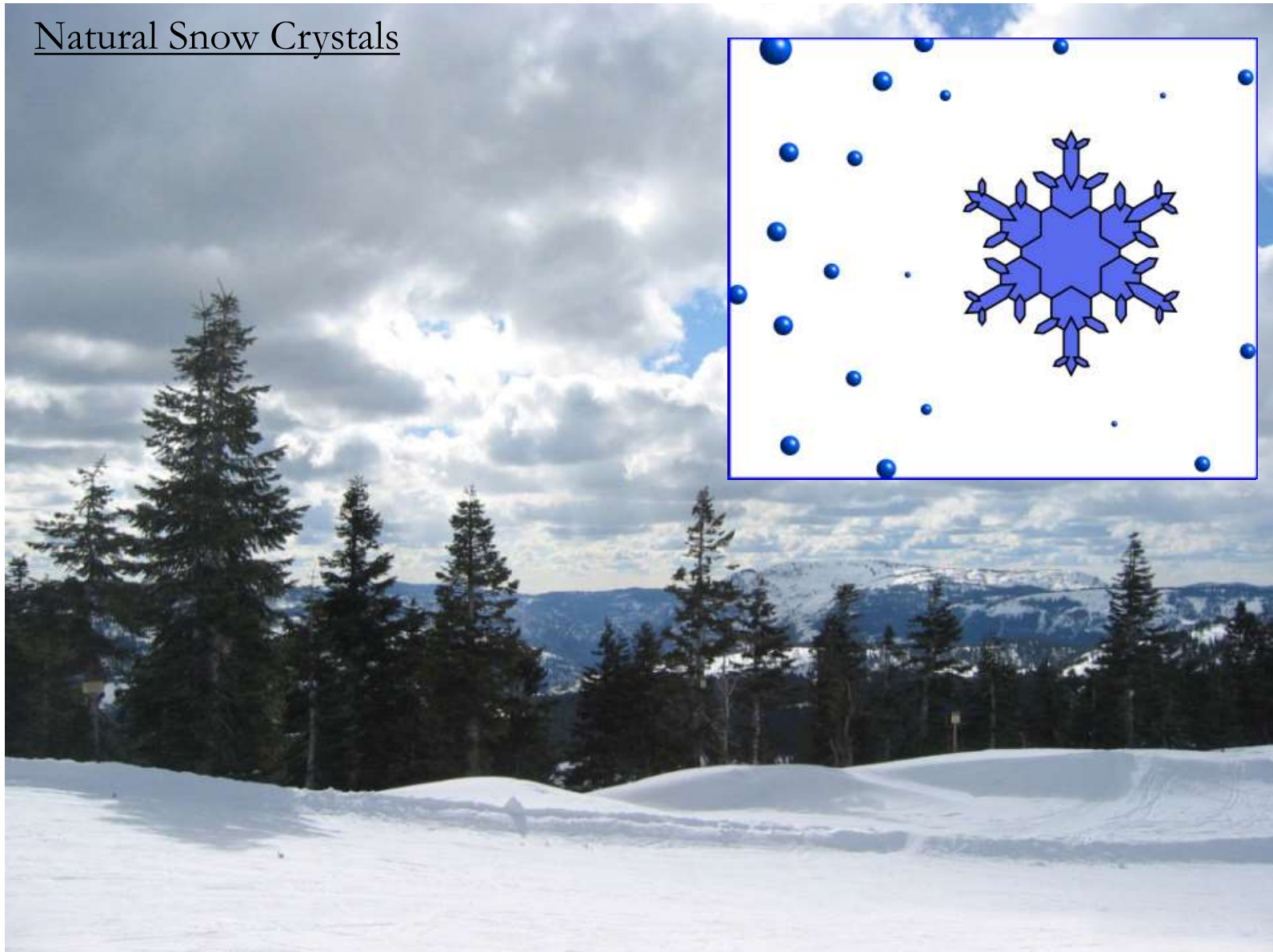
The *Rosetta Stone* for snow crystals...
Based on empirical observations...
Valid in normal air only...



Especially puzzling:
Overall morphology
depends strongly on
temperature

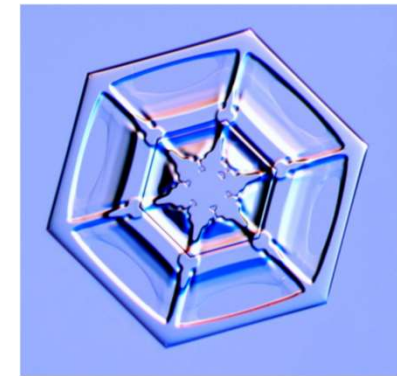
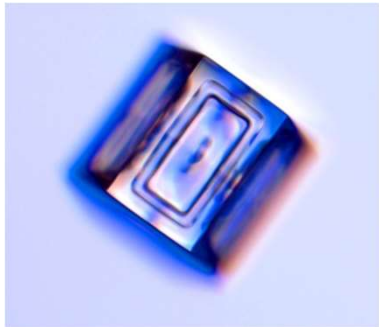


Natural Snow Crystals



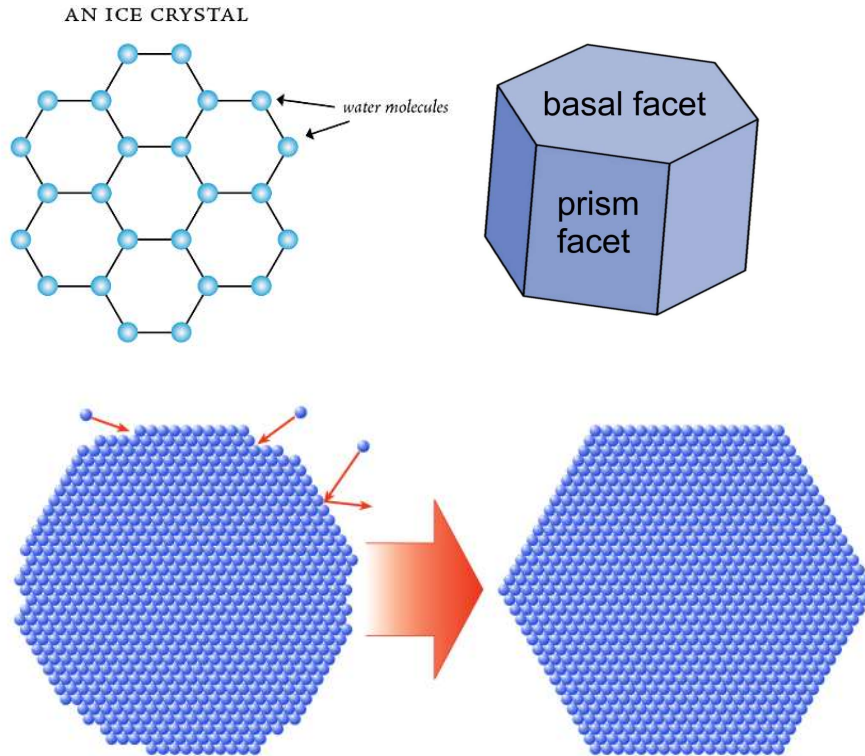
Snowflake Physics I – Attachment Kinetics

Anisotropic attachment kinetics → Facets

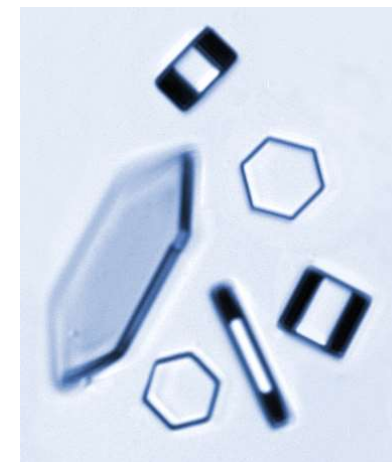
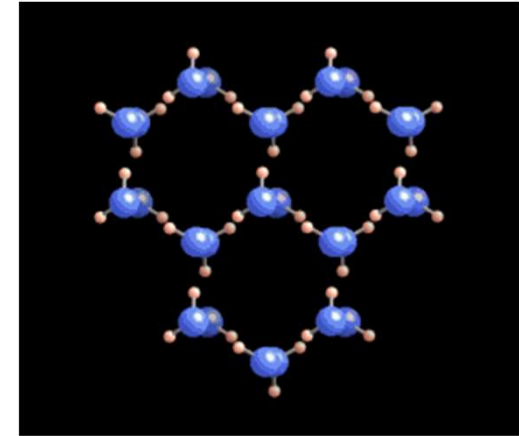


Highly Anisotropic Attachment Kinetics

(Surface energy not so important; nearly isotropic)



Molecules cannot readily attach to smooth surfaces
→ facets form as crystal grows
A non-equilibrium dynamical process



Tiny,
laboratory
grown
snow
crystals
~0.1 mm

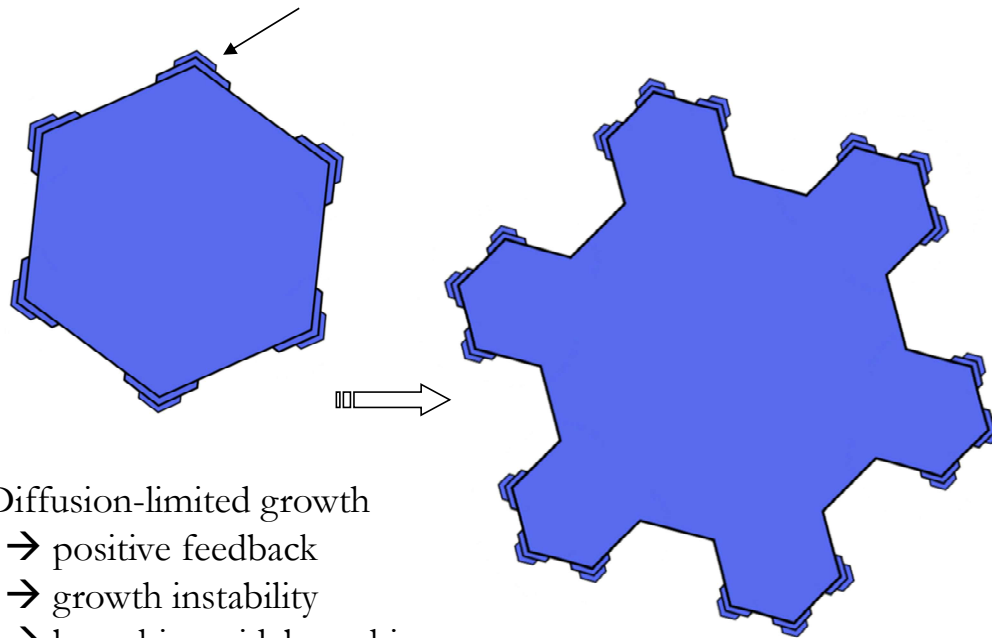
→ no 4-, 5-, 7-, 8-sided snow crystals

Faceting is how the geometry of the water molecule is transferred to the geometry of a crystal.

Snowflake Physics II – Diffusion Limited Growth

Particle diffusion → Branching Instability
(a.k.a. the Mullins-Sekerka instability; 1963)

The six corners stick out farther into the humid air
So the corners grow faster... branches sprout



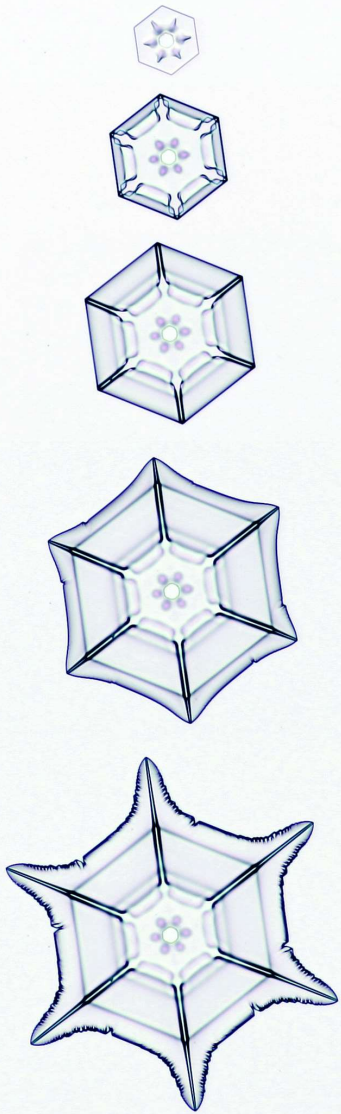
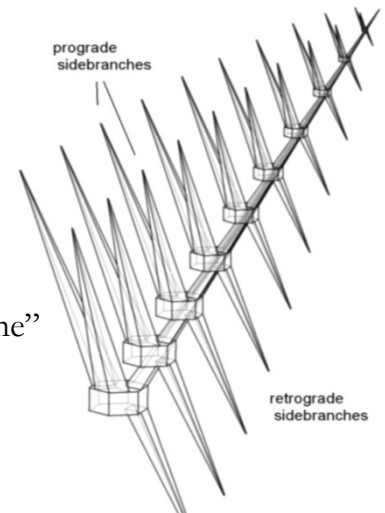
Diffusion-limited growth
→ positive feedback
→ growth instability
→ branching, sidebranching ...

Much scientific literature on diffusion-limited growth

Platelike
dendrite
thin & flat
-15C

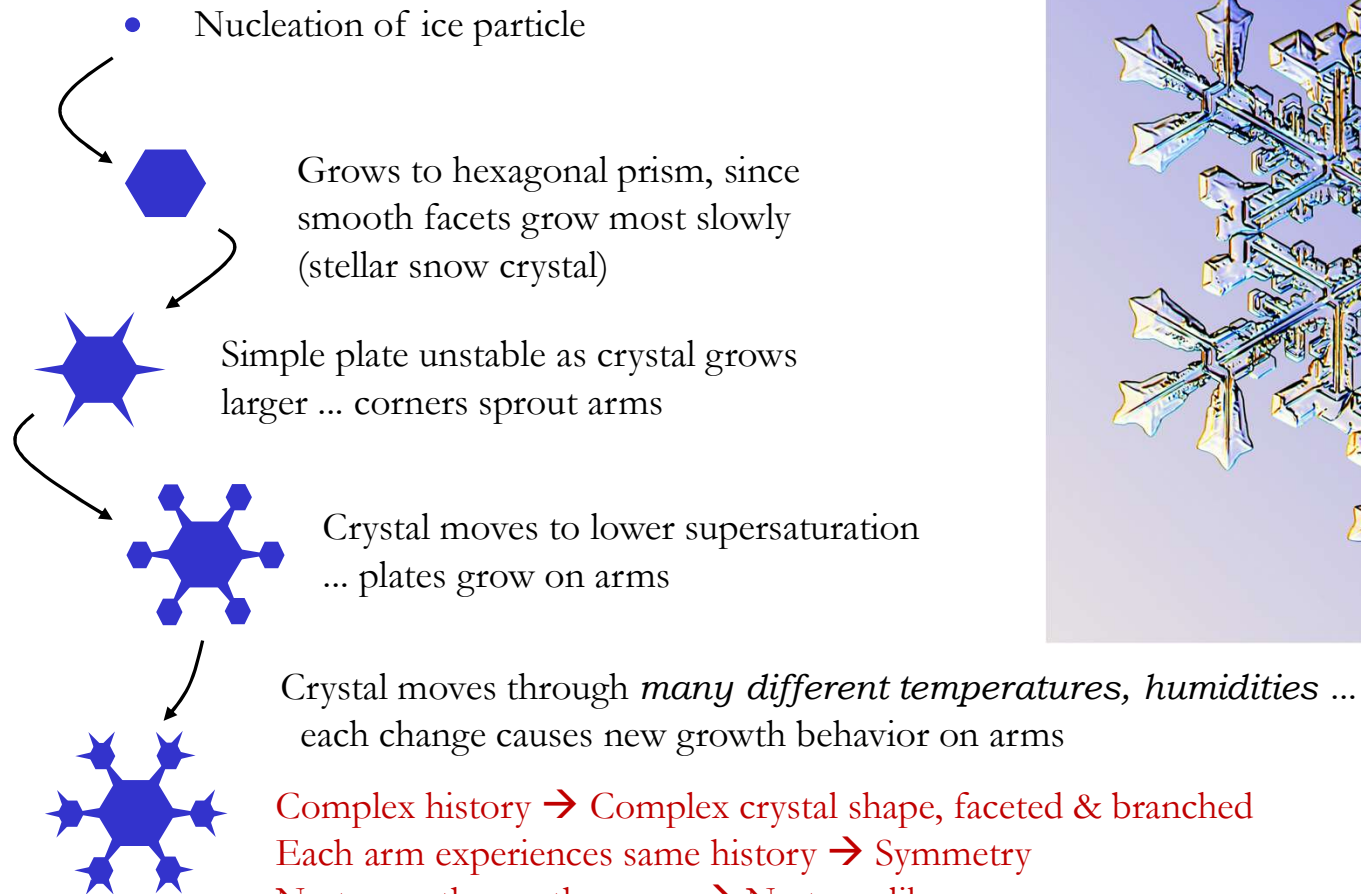


“Fishbone”
dendrite
-5C



Snowflake Physics III - Complexity and Symmetry

(Why snowflakes are symmetrical, and why “no two are alike”)

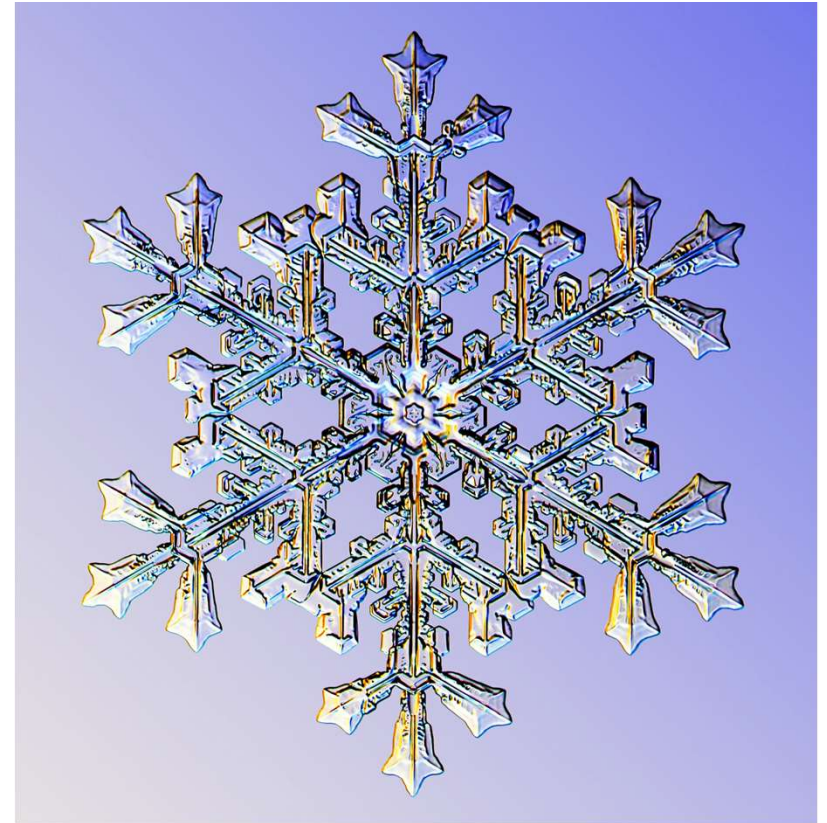


Complex history → Complex crystal shape, faceted & branched

Each arm experiences same history → Symmetry

No two paths are the same → No two alike

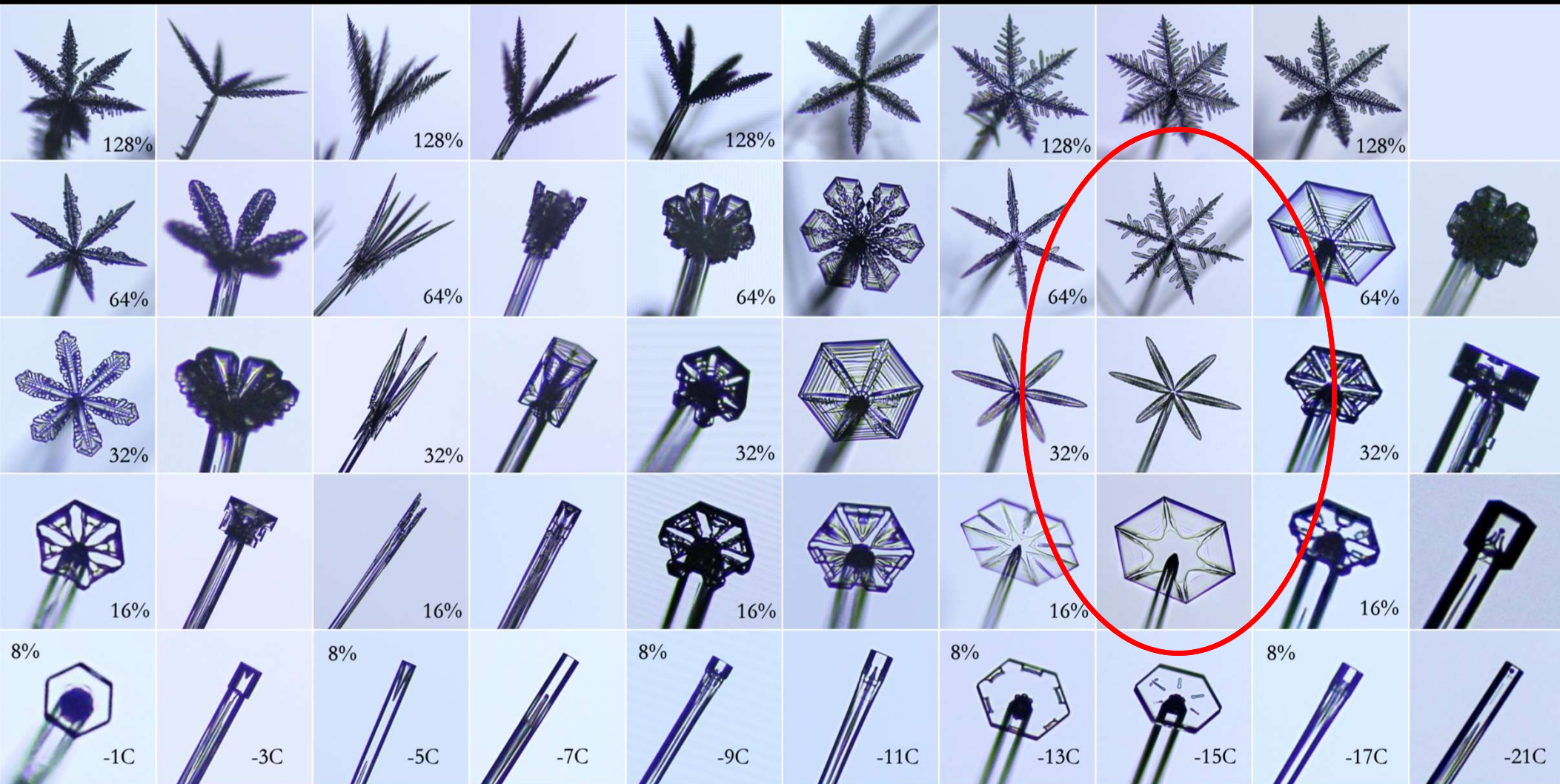
All because growth sensitive to temperature, humidity



T ~ -1 C → Platelike

T ~ -5 C → Columnar

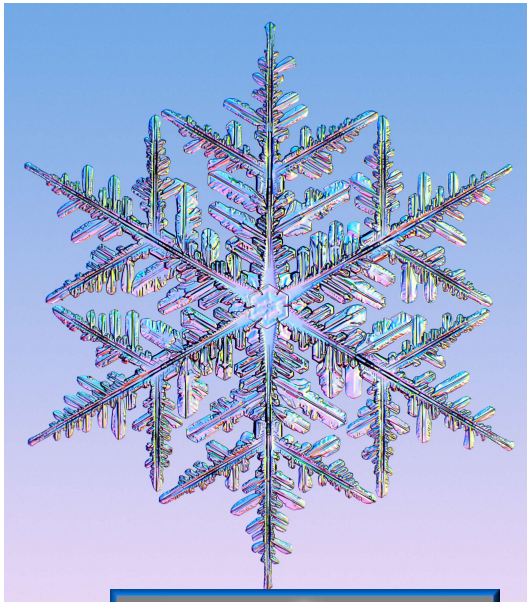
T ~ -15 C → Platelike



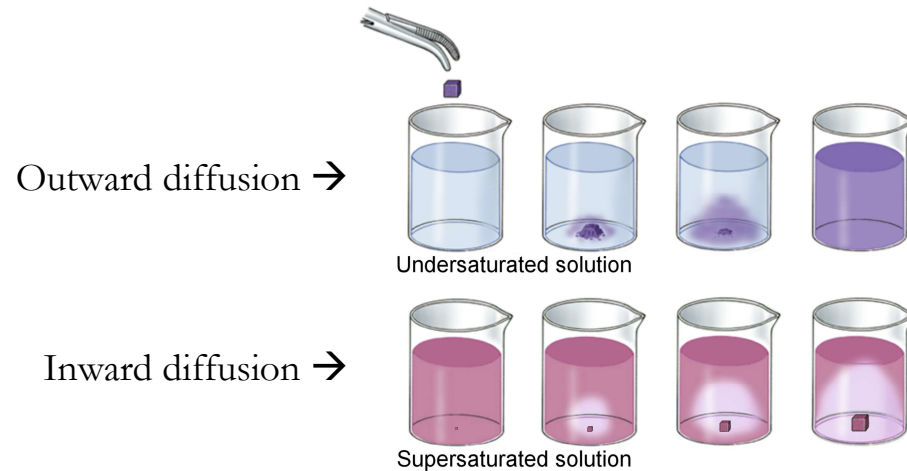
Growth slow → faceting dominates; Fast → branching dominates

... Attachment kinetics on faceted surfaces key factor

DIFFUSION-LIMITED GROWTH OF SNOW CRYSTALS



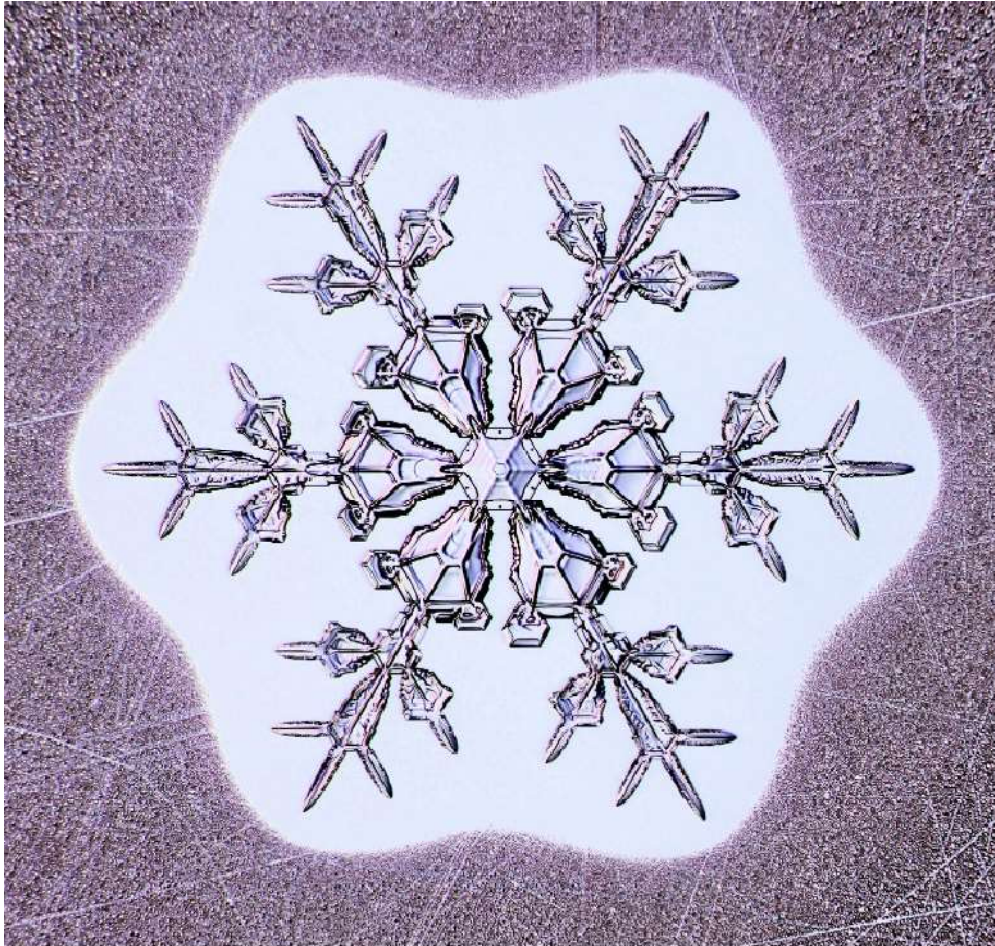
Largely responsible for structure formation, especially branching



Supersaturation depleted around growing crystal → diffusion-limited growth

← See effect in numerical models
Here 2D model; outer shading shows water vapor supersaturation

Janko Gravner and David Griffeath, Modeling snow crystal growth II: A mesoscopic lattice map with plausible dynamics, *Physica D* 237, 385-404, 2008.



← **Laboratory example of water vapor depletion around growing crystal**

>> Snow crystal growing on glass substrate

>> Moist air blows down onto crystal to create supersaturated environment

>> Water droplets condense on glass, except where supersaturation depleted by growing crystal

DIFFUSION-LIMITED GROWTH: THE SPHERICAL SOLUTION

Diffusion equation: $\nabla^2 \sigma = 0$; $\sigma(r)$ = supersaturation field (Laplace approx.)

Attachment kinetics: $v_n = \alpha v_{kin} \sigma_{surf}$; $0 \leq \alpha \leq 1$... attachment coefficient
(Hertz-Knudsen relation – 1882)

Can add: Latent heating + Heat diffusion (double diffusion problem)

Surface energy effects (Gibbs-Thomson effect)

On substrate: Hemispherical solution useful

→ In air:

Mainly **attachment kinetics + particle diffusion**

Thermal effects from latent heating ... minor (not zero)

Surface energy effects ... minor (mostly ignore if $r > 1\mu\text{m}$)

→ In near vacuum:

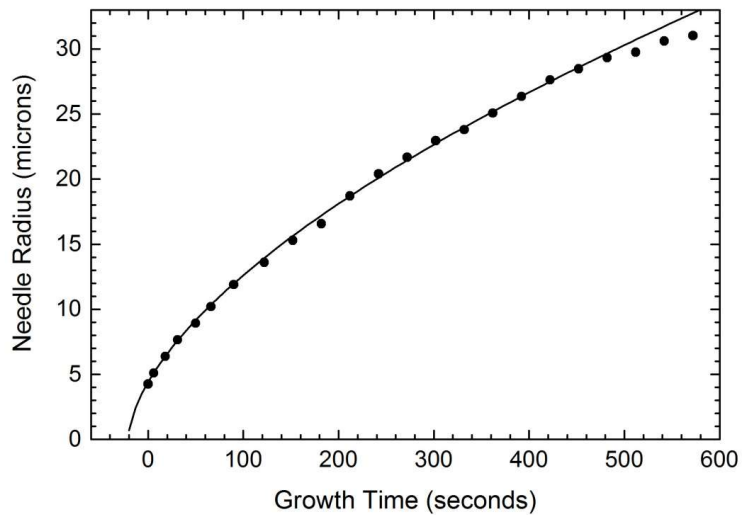
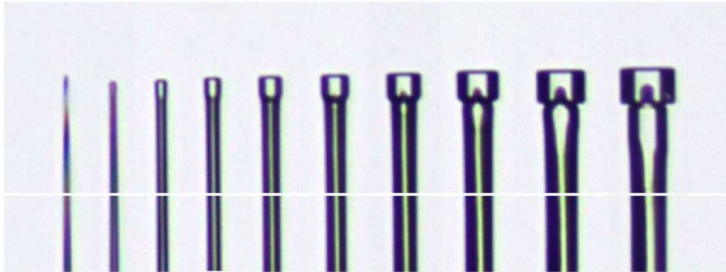
Mainly attachment kinetics + heat diffusion



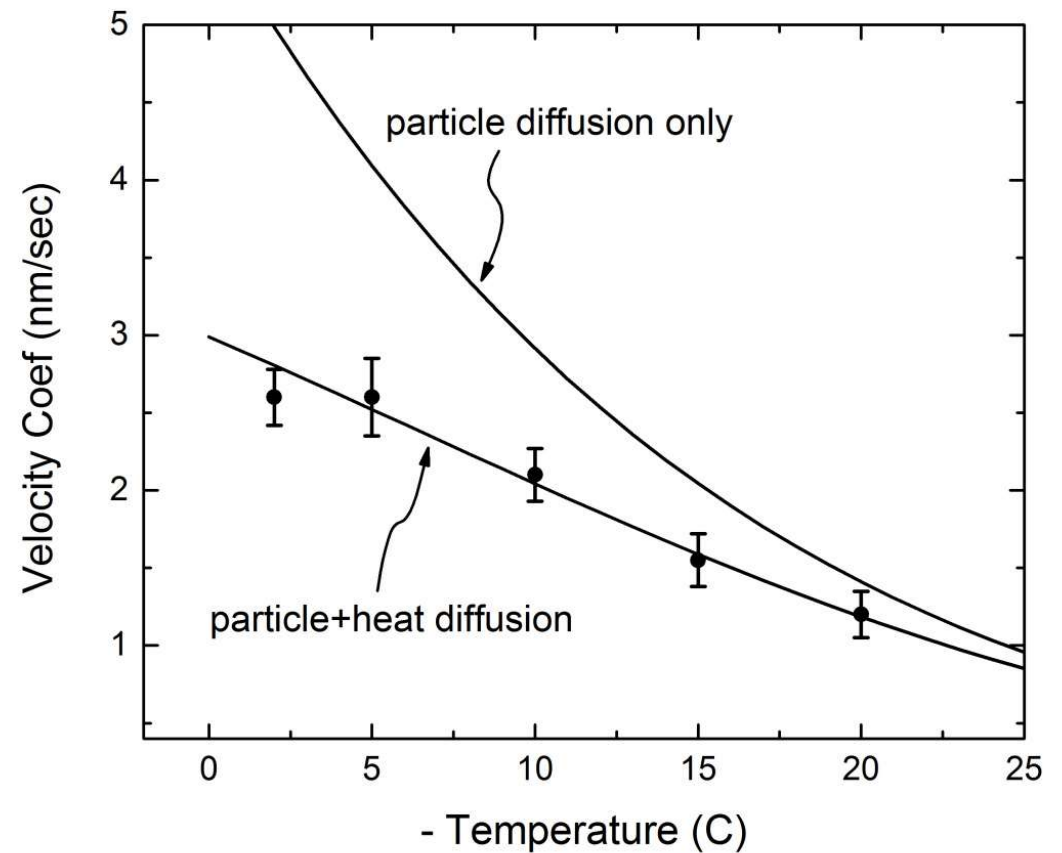
*Spherical solution extremely useful for understanding
the relative important of different physical processes...*

DIFFUSION-LIMITED GROWTH: CYLINDRICAL GROWTH

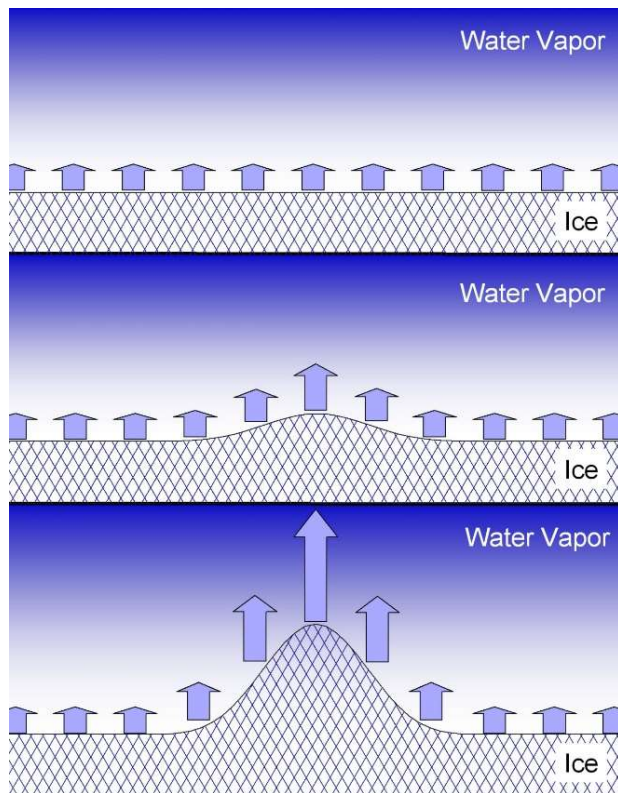
Can solve double-diffusion problem analytically: heat and particle diffusion



Measure as function of (T, σ)



THE MULLINS-SEKERKA INSTABILITY



← Planer (1D) solution of diffusion equation
Crystal grows upward at uniform rate, $v = \text{const}$
Simple 1D supersaturation gradient

← But ... simple 1D solution is not a **stable** solution...

Spherical growth →
Another 1D solution of the
diffusion equation → $v \sim 1/R$

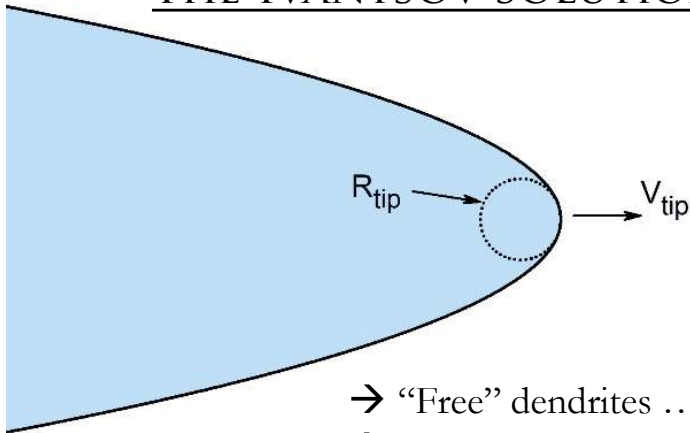
... but, sphere also **unstable**
to the Mullins-Sekerka instability



W. W. Mullins and R. F. Sekerka, Stability of a planar interface during solidification of dilute binary alloy, J. Appl. Phys 35, 444-451, 1964.

THE IVANTSOV SOLUTION

G. P. Ivantsov, Dokl. Akad. Nank USSR 58, 1113, 1947.



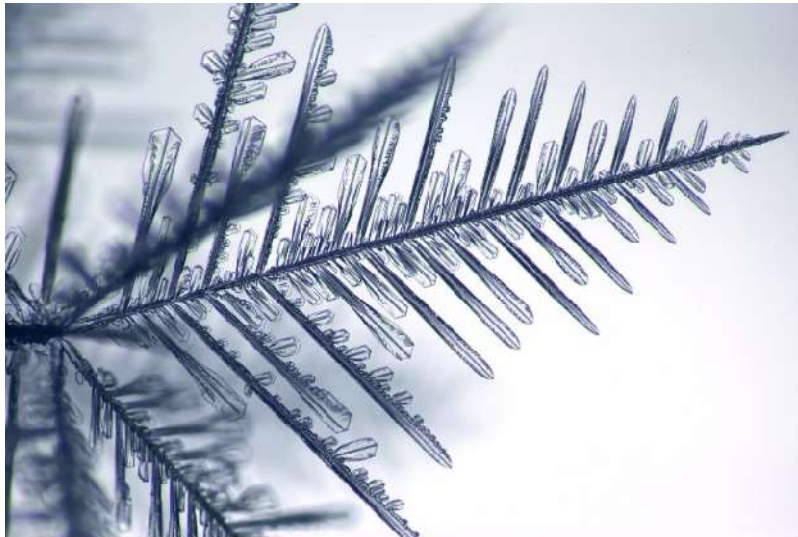
← 2D solution in parabolic coordinates

Paraboloid of revolution

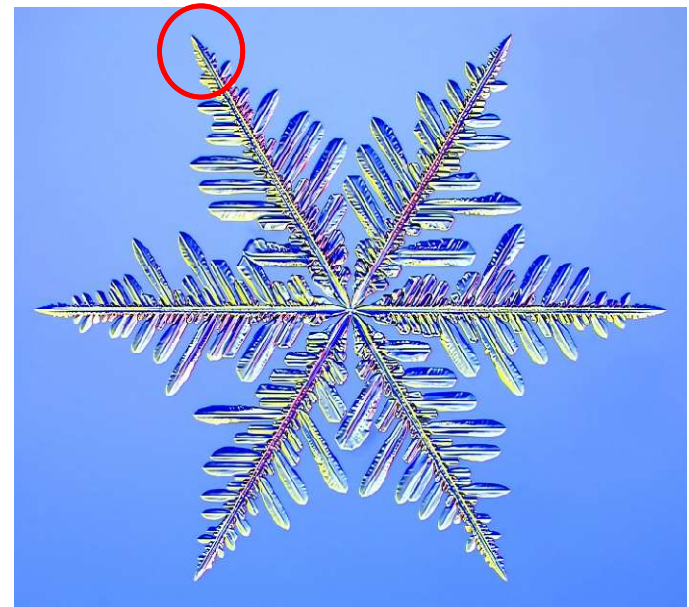
$$v_{tip} = \text{const} \sim 1/R_{tip} \quad (R_{tip} = \text{const})$$

Also **unstable** to the Mullins-Sekerka instability

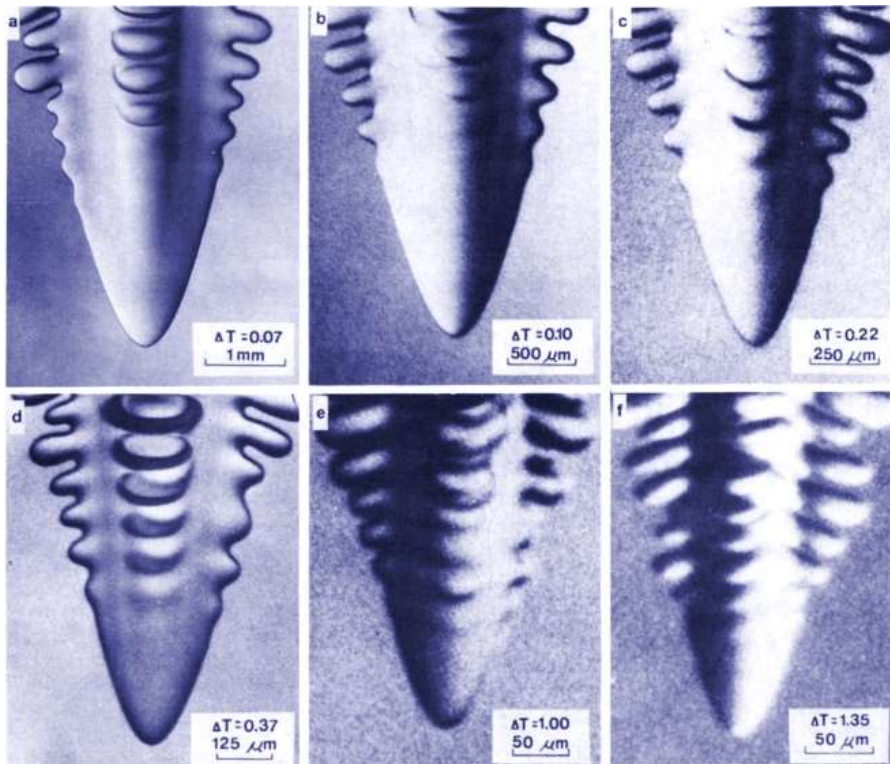
- “Free” dendrites ... widely studied in 1980s
- “Solvability” theory of free-dendrite growth



Focus on tip → ignore sidebranching



THE SELECTION PROBLEM



S.-C. Huang and M. E. Glicksman, Overview 12: Fundamentals of dendritic solidification – II Development of sidebranch structure, Acta Metall. 29, 717-734, 1981.

← Free dendrites growing during the solidification of liquid succinonitrile (a clear, waxy material that melts at 57 C)

Parabolic tip stable in moving frame....

Low supercooling → low v_{tip} , large R_{tip}

High supercooling → high v_{tip} , small R_{tip}

Ivantsov solution says

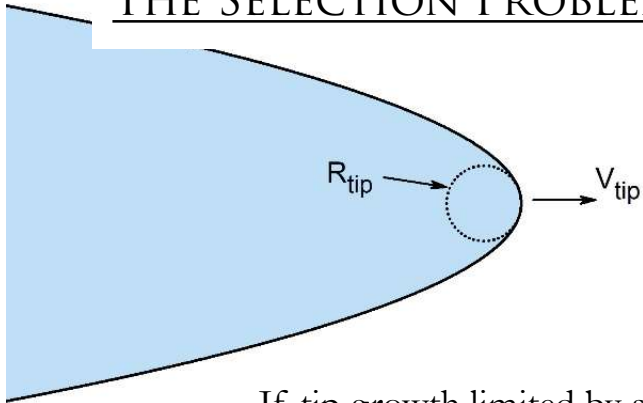
$$v_{tip} = \text{const} \sim 1/R_{tip} \quad (R_{tip} = \text{const})$$

... a *family* of solutions

What “selects” the experimental solution?

Diffusion equation alone is not enough....

THE SELECTION PROBLEM – GROWTH FROM LIQUID AND VAPOR



Spherical solution of diffusion equation
with attachment kinetics &
with surface energy (Gibbs-Thomson)...

$$v \approx \frac{X_0}{R} \left(1 - \frac{X_0}{\alpha R} \right) \left(1 - \frac{d}{\sigma_\infty R} \right)$$

If tip growth limited by surface energy, then solvability theory gives:

$$v_{tip} \sim \sigma_\infty^2; \quad R \sim \frac{1}{\sigma_\infty}$$

Holds for growth from liquid (mostly) \rightarrow
(as function of supercooling)

If tip growth limited by attachment kinetics, then

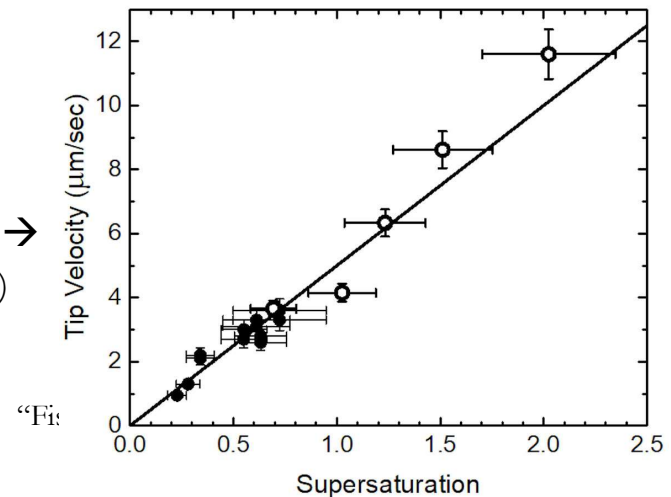
$$v_{tip} \sim \sigma_\infty; \quad R \sim \text{const}$$

Holds for growth from vapor (mostly) \rightarrow
(as function of vapor supersaturation)

$0 \leq \alpha \leq 1$... attachment coefficient

d ... Gibbs-Thomson length = 1 nm

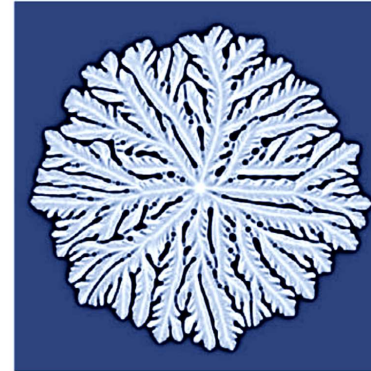
Pond ice, Bathsheba Grossman



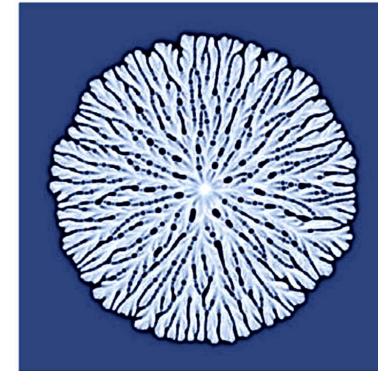
THE SELECTION PROBLEM – ANISOTROPY NEEDED

Computer simulations showed Ivantsov solutions **unstable** to “tip splitting” !

← Greater anisotropy



Less anisotropy →



Overarching conclusions from solvability theory:

- Stable parabolic tip requires crystal anisotropy
- Surface energy usually dominates in growth from melt
- Attachment kinetics usually dominates in growth from vapor
- Need full 3D computer simulations to make further progress...

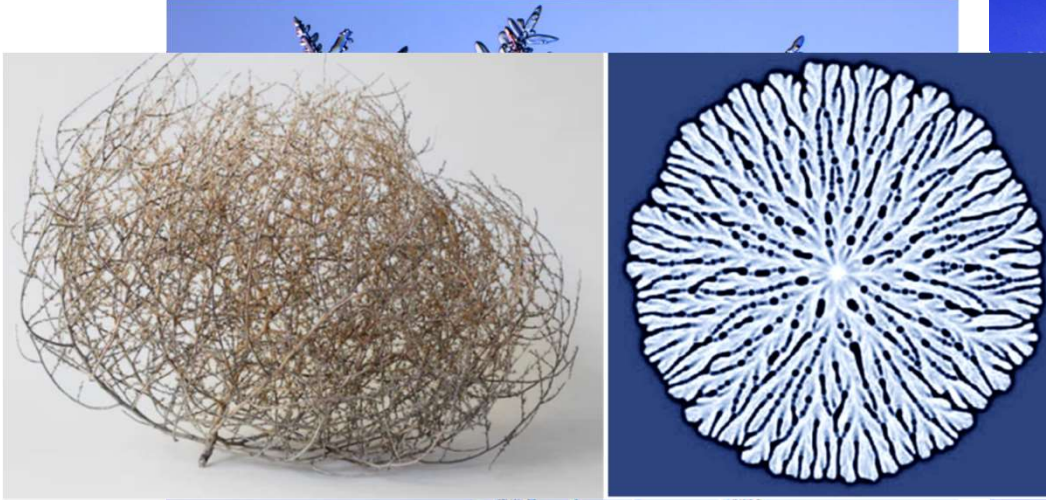
Facets → Anisotropic attachment kinetics (not surface energy)

Tip splitting in stellar dendrite growth at high σ_∞ →



EXAMPLES OF DIFFUSION-LIMITED SNOW CRYSTAL GROWTH

1. Fernlike Stellar Dendrites



Rapid growth, erratic sidebranching

Platelike form **not** explained by Mullins-Sekerka instability...
Large-scale morphological asymmetry indicates
anisotropic attachment kinetics.

(Isotropic → “tumbleweed” growth)



10mm from tip to tip, self-similar

ORDER AND CHAOS

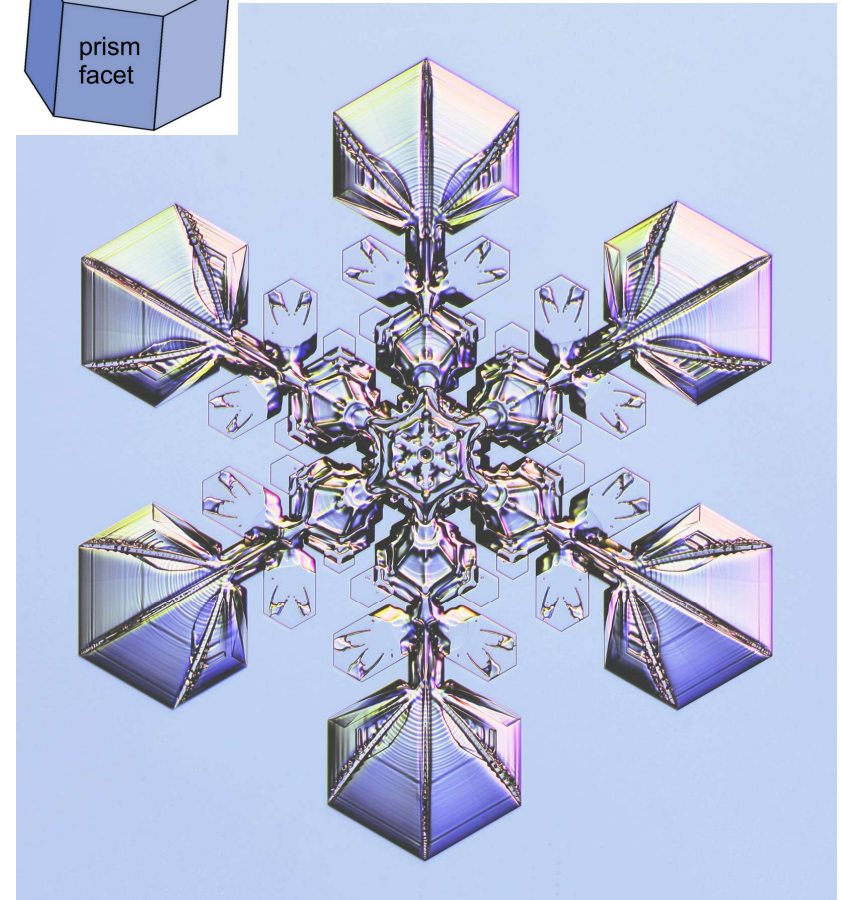
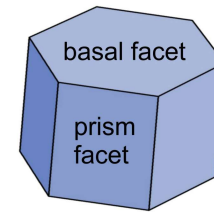
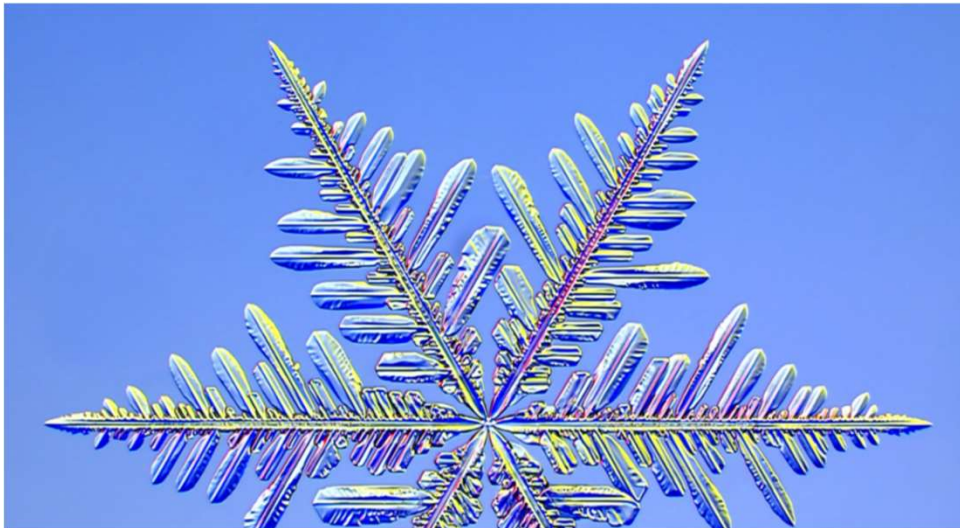
Large-scale morphology determined by attachment kinetics
Slow growth → faceting -- order

Complexity arises from diffusion-limited growth
Fast growth → branching -- chaos

>> Must have **large-scale anisotropy in attachment kinetics**
to make thin plates or slender needles

>> Surface energy nearly isotropic (ice does not cleave in facets)

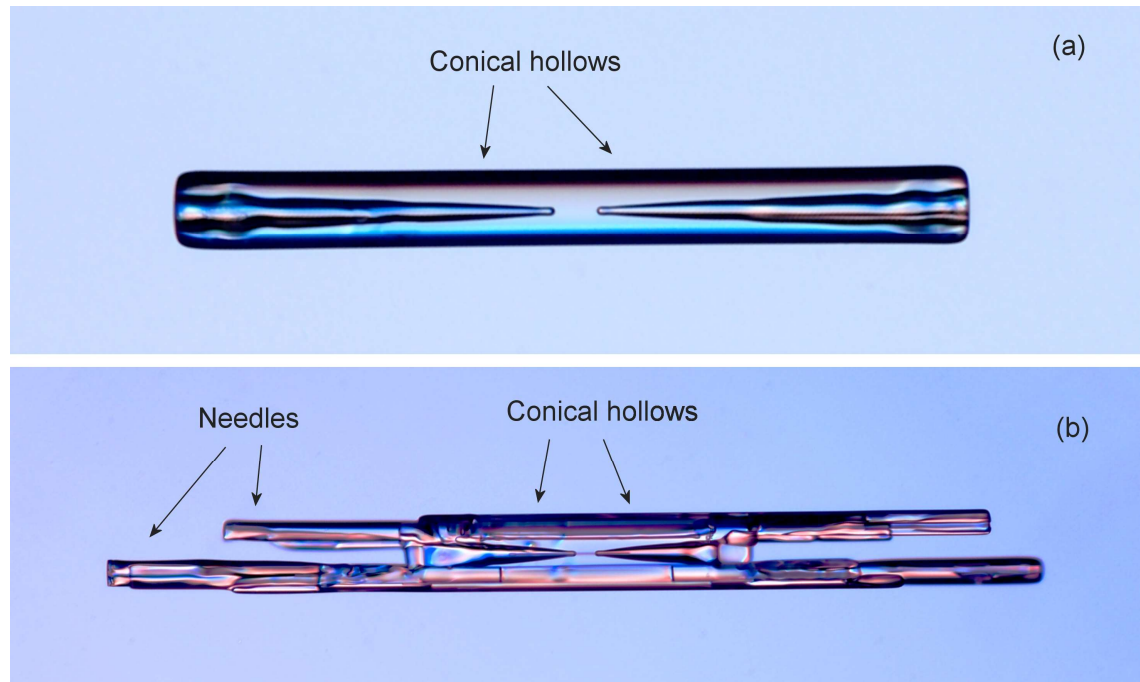
>> In crystal growth, facets \leftrightarrow anisotropic attachment kinetics



Interplay of faceting and branching
Not grown under constant conditions

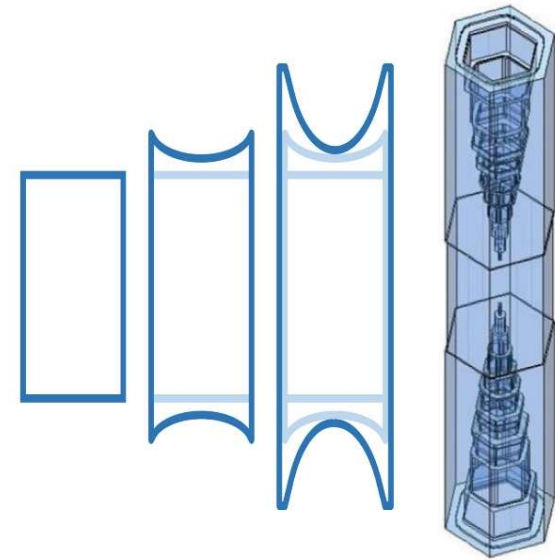
EXAMPLES OF DIFFUSION-LIMITED SNOW CRYSTAL GROWTH

2. Hollow Columns and Needle Clusters



Columnar form **not** explained by Mullins-Sekerka instability...
Large-scale asymmetry indicates anisotropic attachment kinetics.

Basal faces unstable to hollowing:

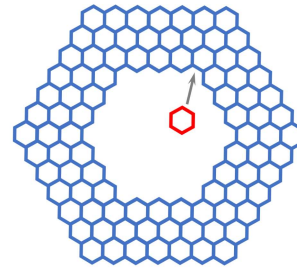


Model from: Janko Gravner and David Griffeath, Modeling snow-crystal growth: A three-dimensional mesoscopic approach, Phys. Rev. E79, 011601, 2009.

EXAMPLES OF DIFFUSION-LIMITED SNOW CRYSTAL GROWTH

3. Ice bubbles

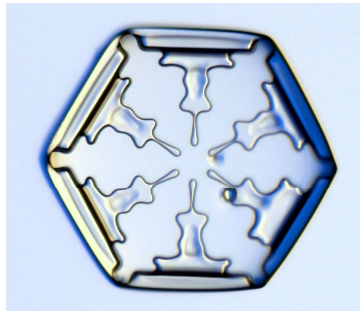
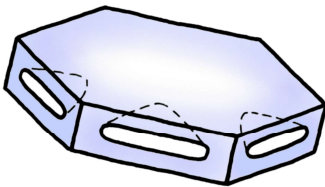
In columns:



Exterior growth slow
Interior growth fast...
Fast → Hollow column
Then slow → Bubble

In plates:

Hollow plate forms...



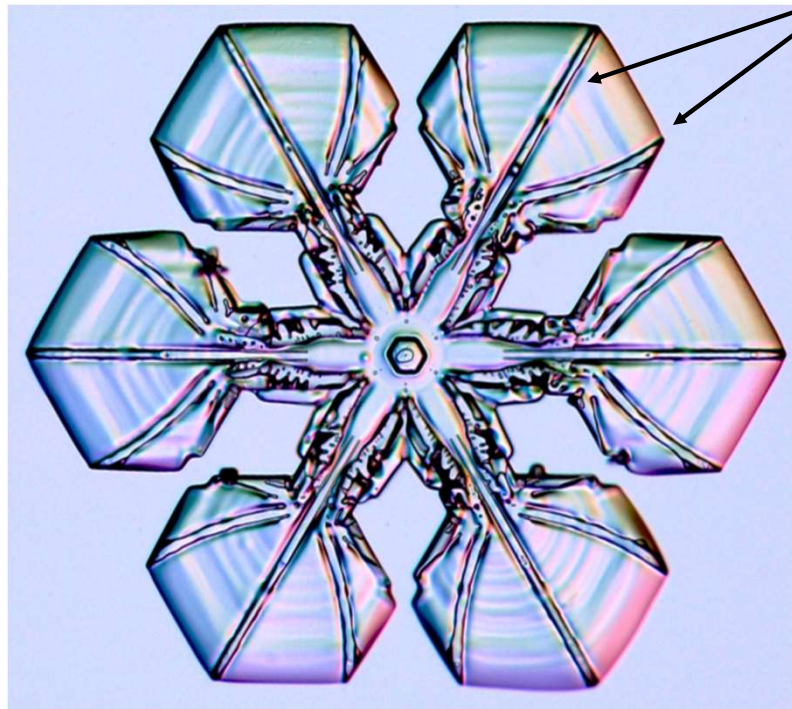
Hollows → bubbles

Interference colors
→ Flat bubbles
... just a few μm thick
... interior faceted

Don Komarechka

EXAMPLES OF DIFFUSION-LIMITED SNOW CRYSTAL GROWTH

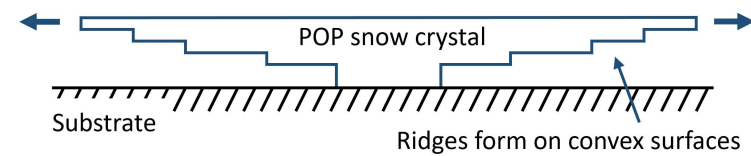
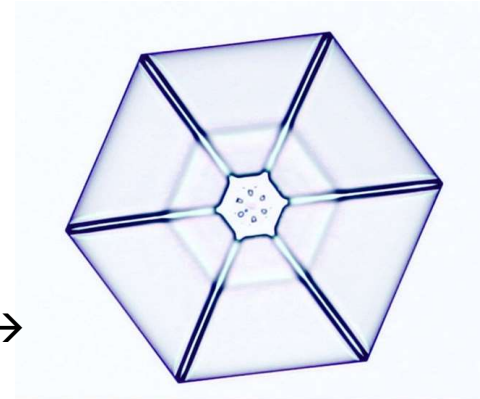
4. Ridge structures



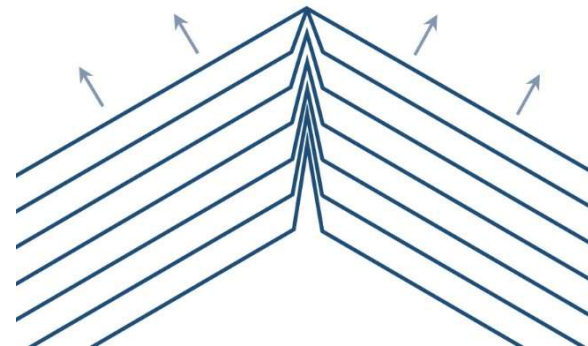
Ridges

← natural

lab-grown →



A 2D Mullins-Sekerka instability...
Step growth unstable to branching...



“ELECTRIC” ICE NEEDLES (AN APPLICATION OF SOLVABILITY THEORY)



Vapor diffusion chamber
→ High supersaturation
(in air at 1 atm)

Insert wire (bottom)
→ Covers with frost

Apply +2000 volts DC
→ “Electric” needles grow

Chemical coaxing
→ c-axis needles

Remove high voltage
Move to 2nd diffusion
chamber
→ Normal growth

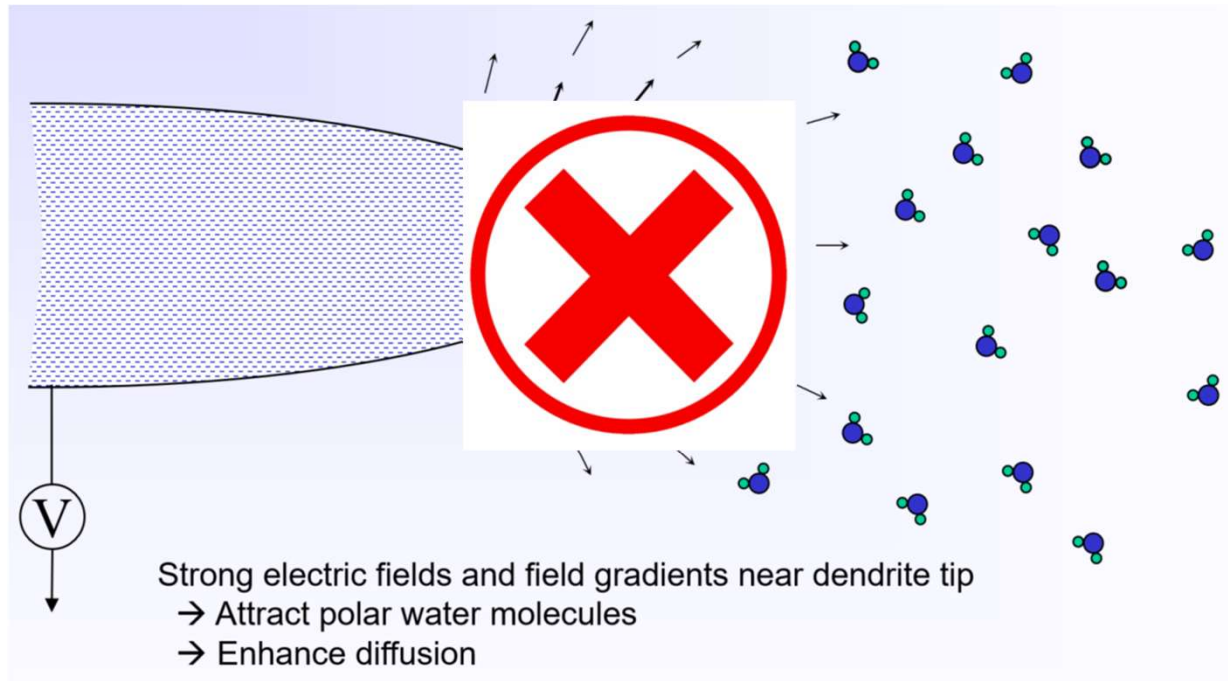


First e-needles: J. T. Bartlett, A. P. van den Heuvel, and B. J. Mason, Growth of ice crystals in an electric field, *Zeit. Fur Ange. Math. Phys.* 14, 599-610, 1963.

Theory, c-axis e-needles: K. G. Libbrecht and T. Crosby and M. Swanson, Electrically enhanced free dendrite growth in polar and non-polar systems, *J. Cryst. Growth* 240, 241-254, 2002.

“ELECTRIC” ICE NEEDLES - THEORY

The obvious (but wrong) explanation:
electrically enhanced diffusion of polar molecules



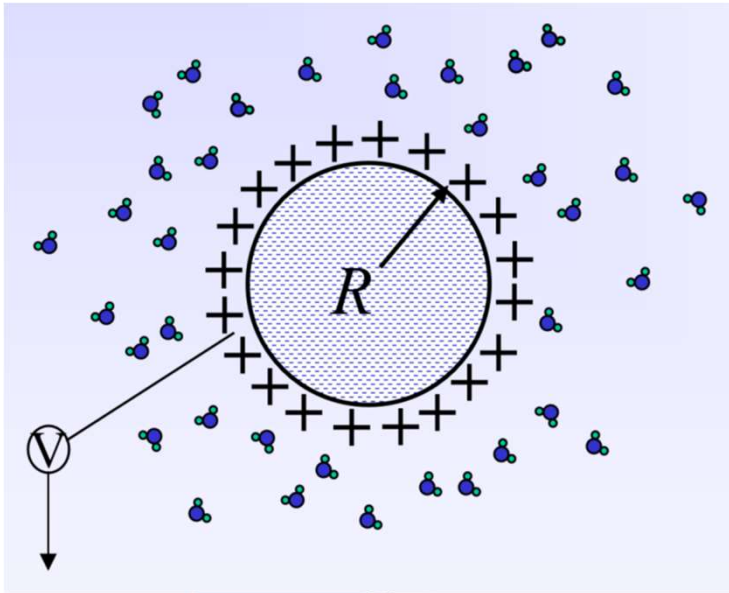
Why wrong?

Polar water molecules have lower energy in strong E fields

→ Tip vapor pressure higher than normal ($E = 0$ or $P = 0$)

→ Effects cancel!

“ELECTRIC” ICE NEEDLES - THEORY



The correct explanation:

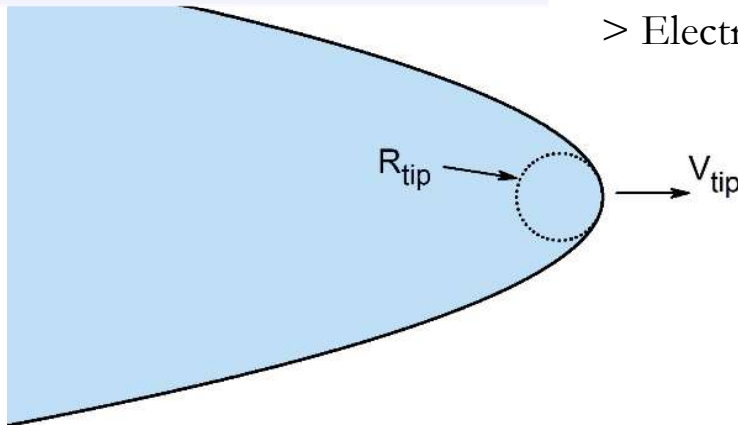
Electrostatic effect **decreases** vapor pressure of charged sphere (independent of polarization)

Vapor pressure of tip changed by:

- > Gibbs-Thomson effect (surface energy)
- > Polarizability effect (for polar molecules)
- > Electrostatics (for polar or nonpolar molecule)

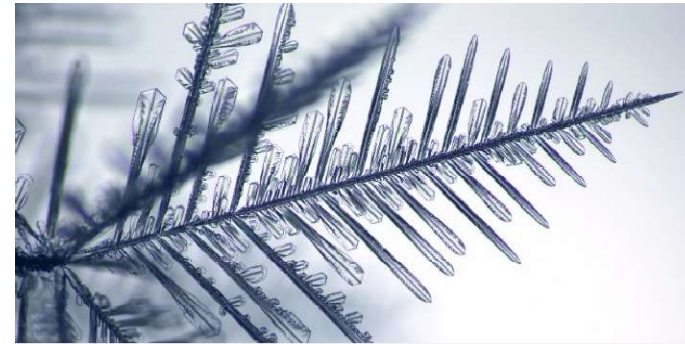
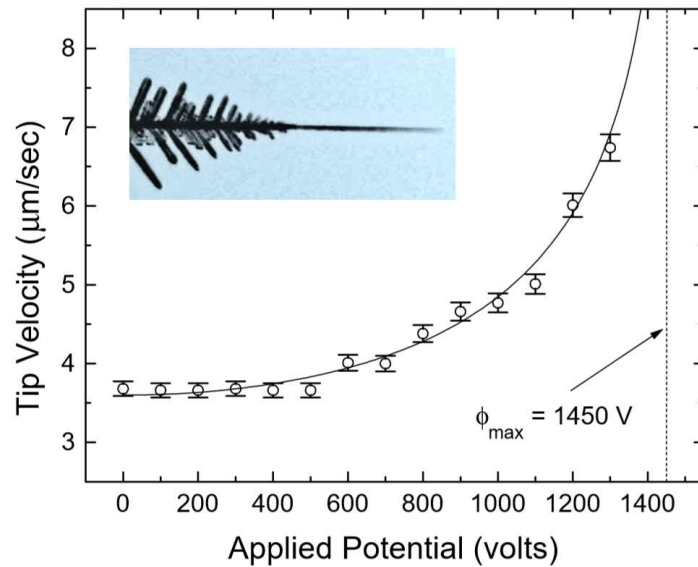
Combine with:

- > Solvability theory
- > Electrically driven diffusion (for polar molecules)

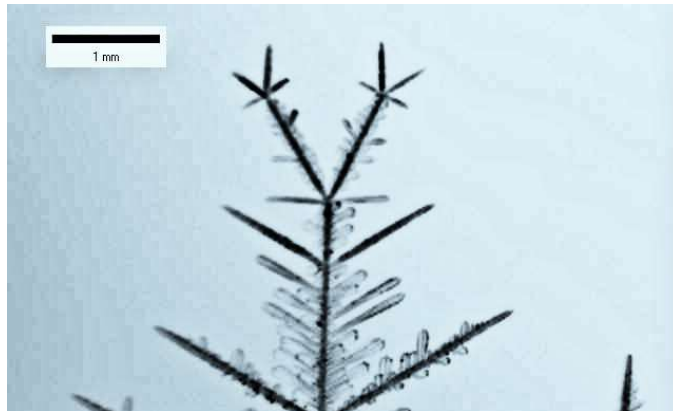


... Messy, but doable with sufficient approximations

“ELECTRIC” ICE NEEDLES – AN ELECTRICALLY DRIVEN TIP-GROWTH INSTABILITY



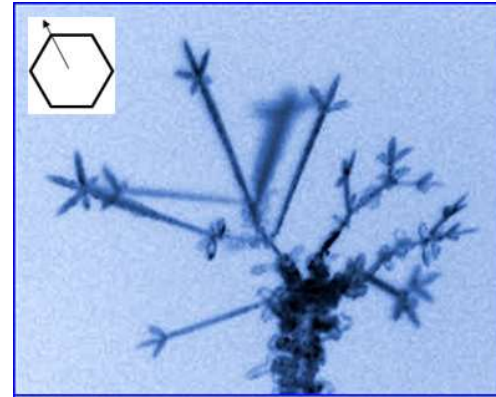
- > Start with a normal snow-crystal dendrite growing on wire
- > Apply +V to wire base
can assume ice \sim conductor, because no current flow
- > At low V, solvability theory \rightarrow enhanced growth
- > When $v_{\text{tip}} \approx 2v_{\text{tip},0} \rightarrow$ theory predicts instability
 $\rightarrow v_{\text{tip}} \rightarrow \infty$
- > Consistent with observations (curve through data above)
- > $V_{\text{applied}} > V_{\text{threshold}} \rightarrow$ e-needle or other phenomena



A SERENDIPITOUS DISCOVERY...



Victoria Tanusheva in the lab, 1997



E-needle
growth axis
unpredictable



is...

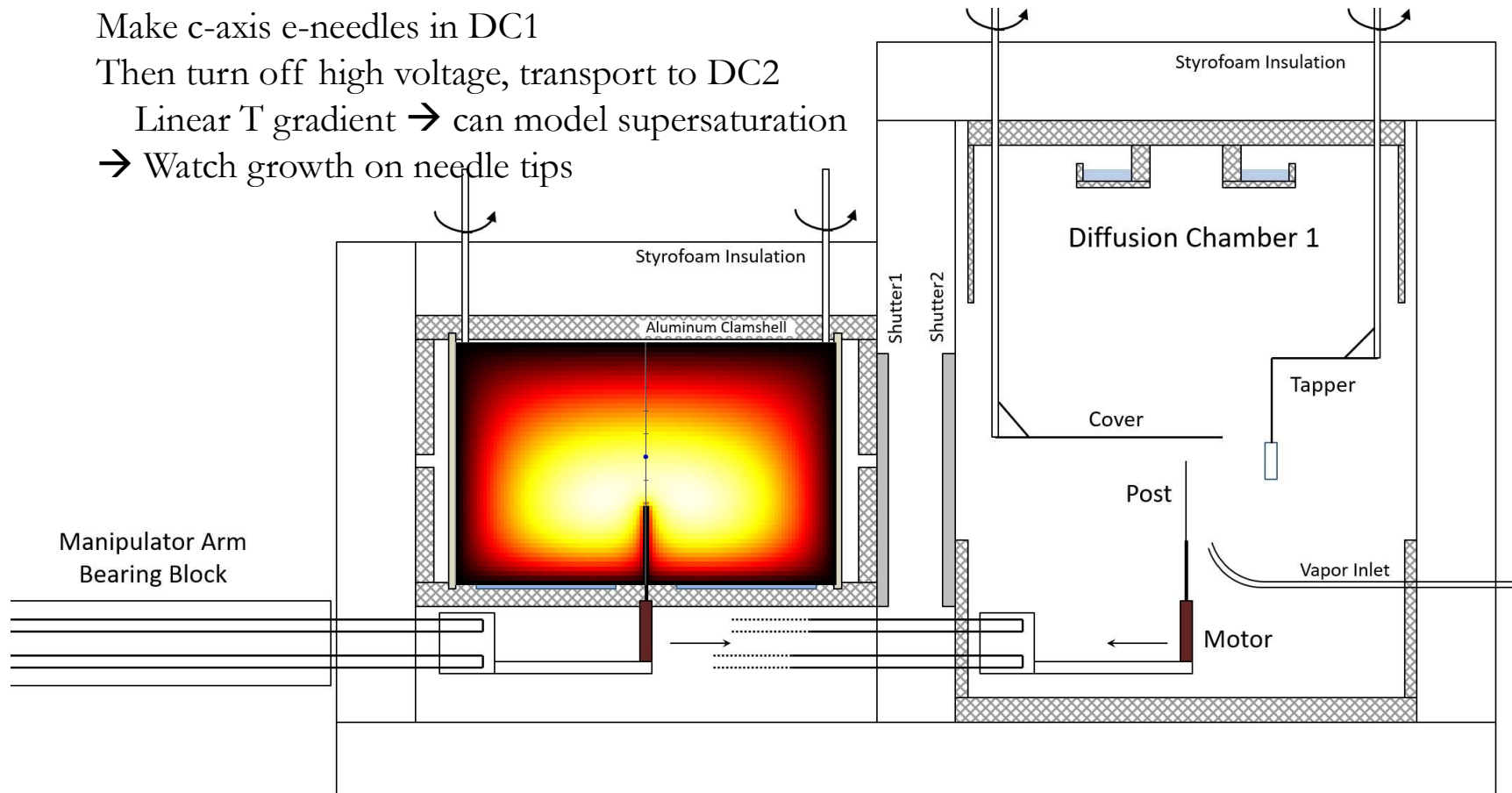
“ELECTRIC” ICE NEEDLES AS A RESEARCH TOOL

Make c-axis e-needles in DC1

Then turn off high voltage, transport to DC2

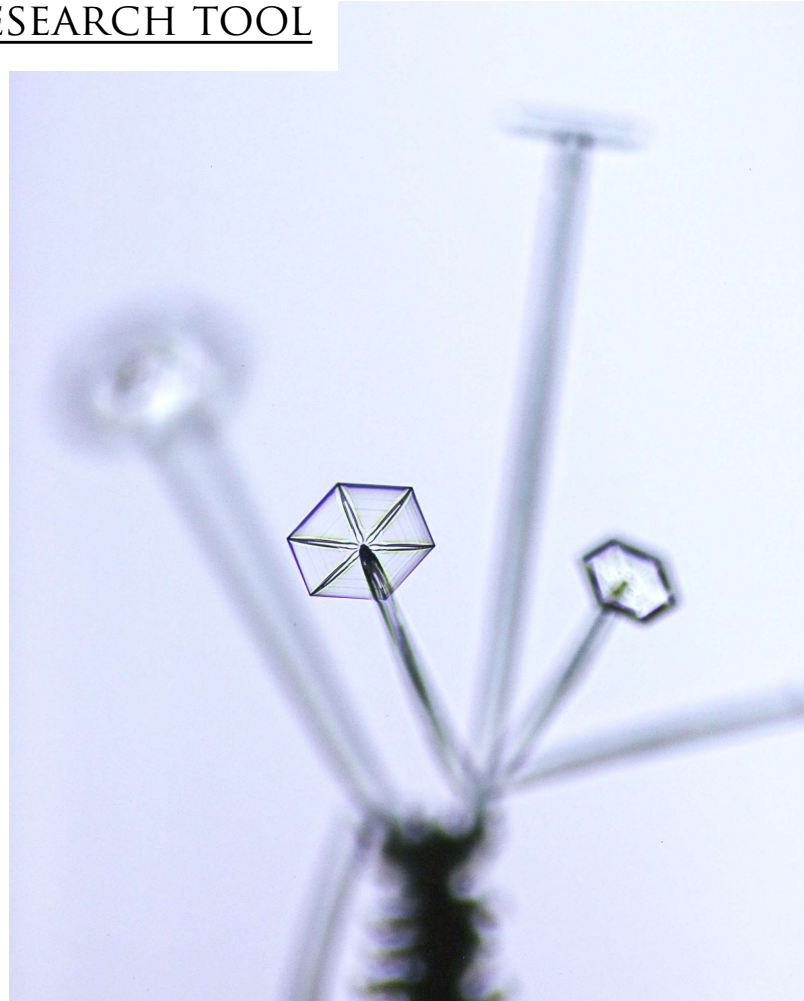
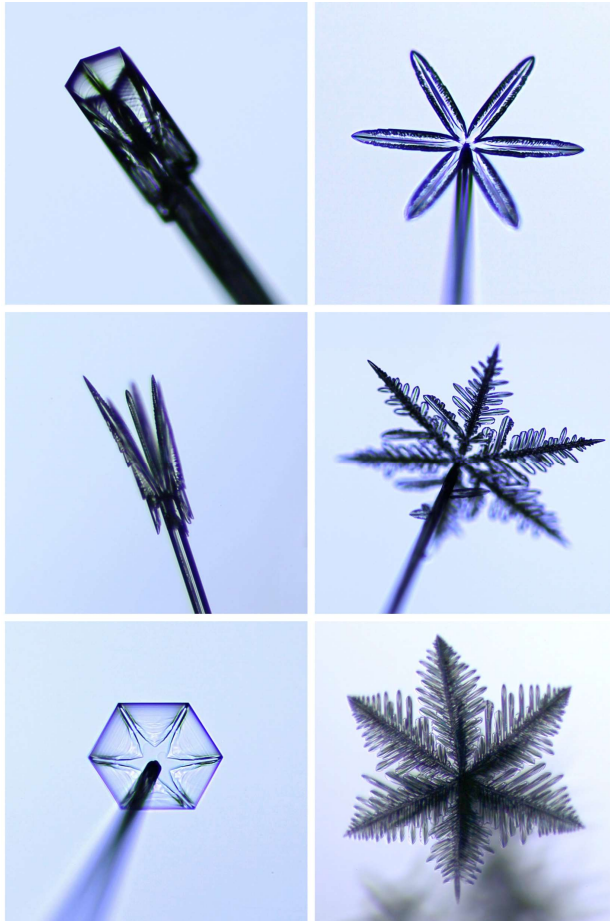
Linear T gradient → can model supersaturation

→ Watch growth on needle tips



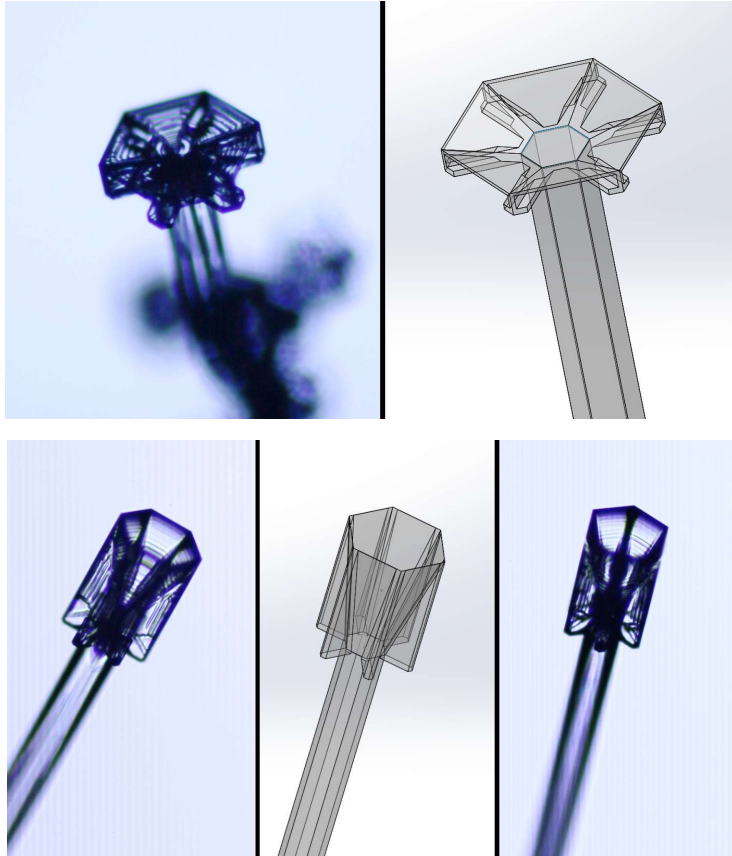
“ELECTRIC” ICE NEEDLES AS A RESEARCH TOOL

Observe a variety of structures

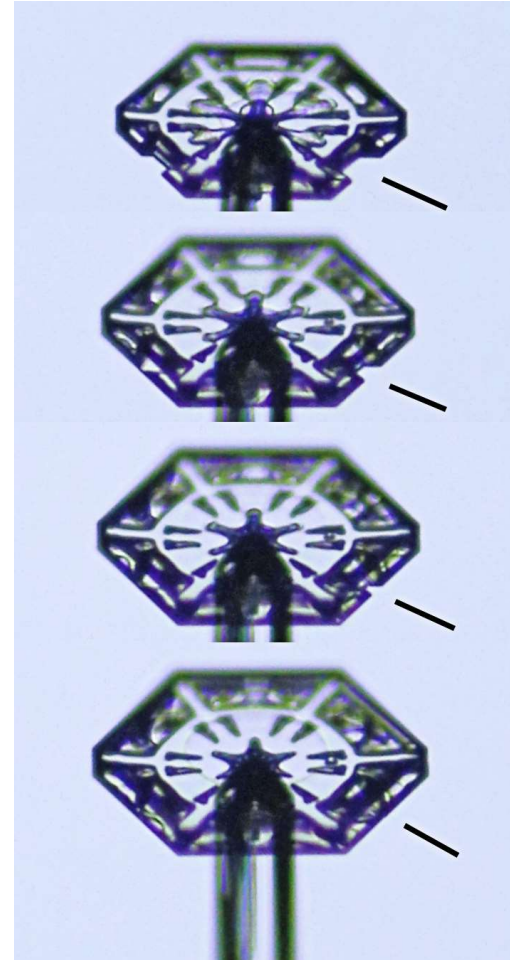


“ELECTRIC” ICE NEEDLES AS A RESEARCH TOOL

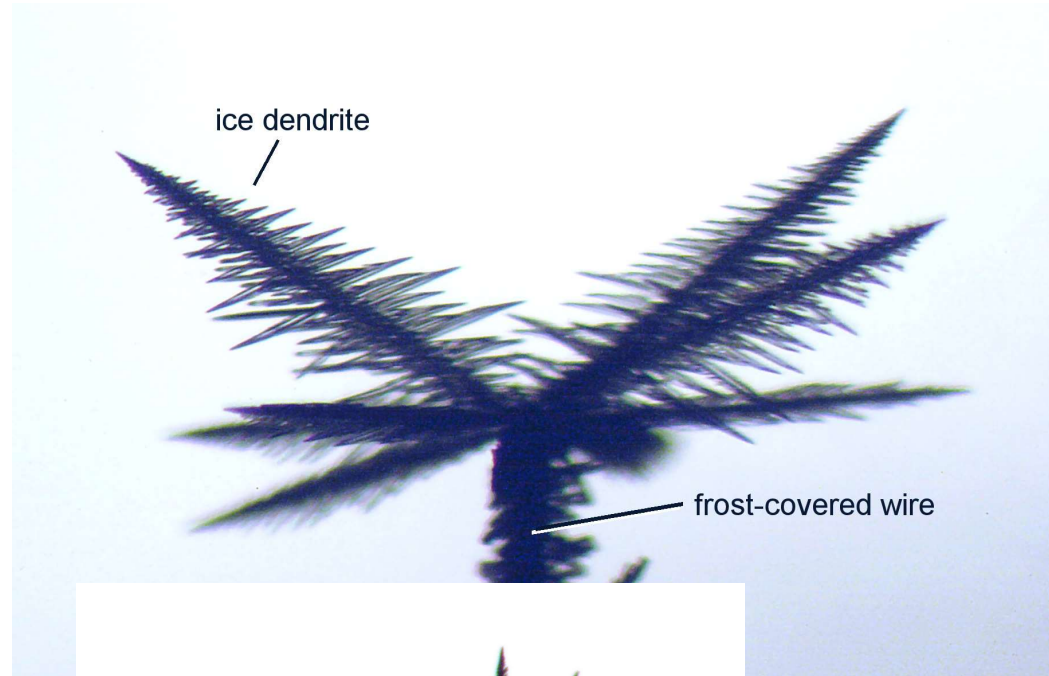
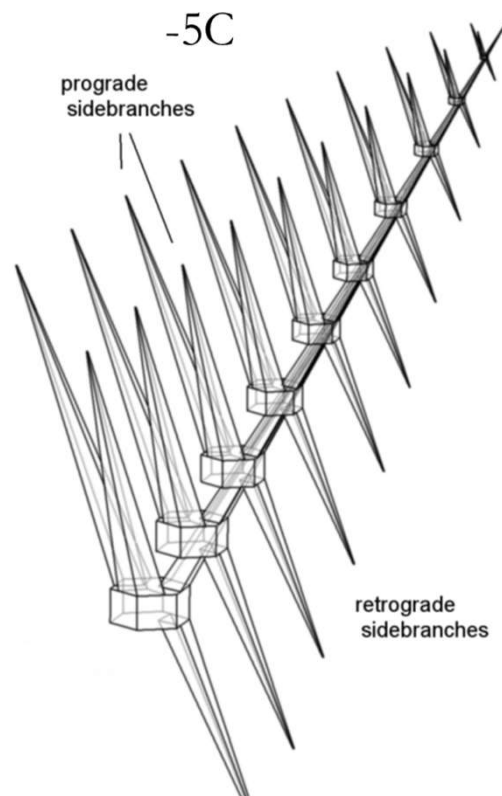
3D Morphology models



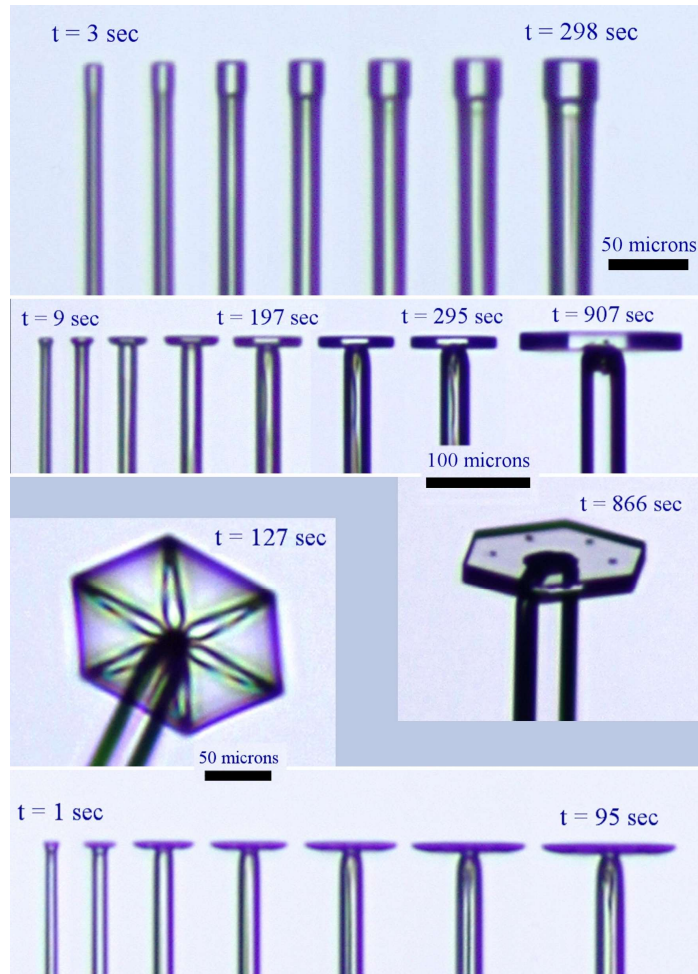
Growth dynamics



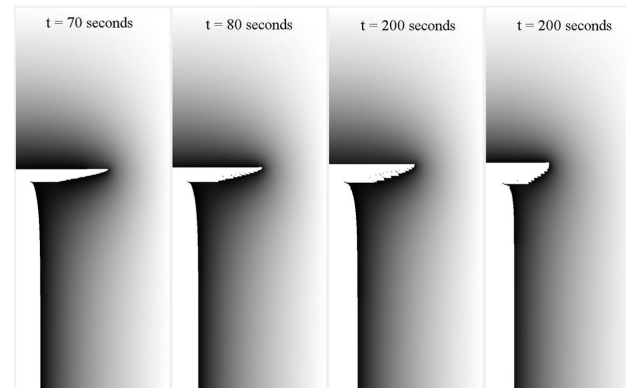
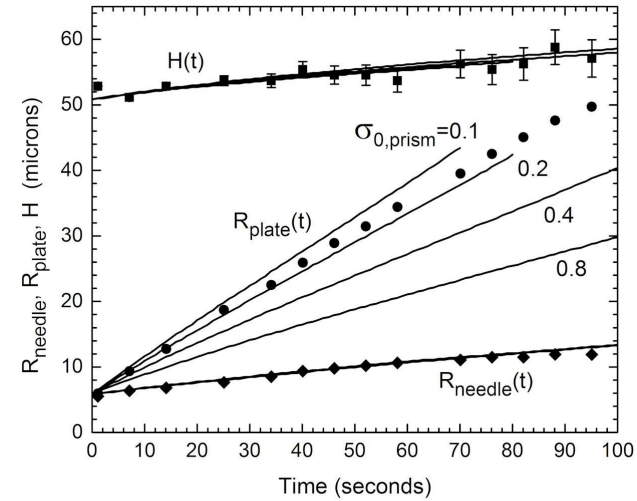
“FISHBONE” DENDRITES



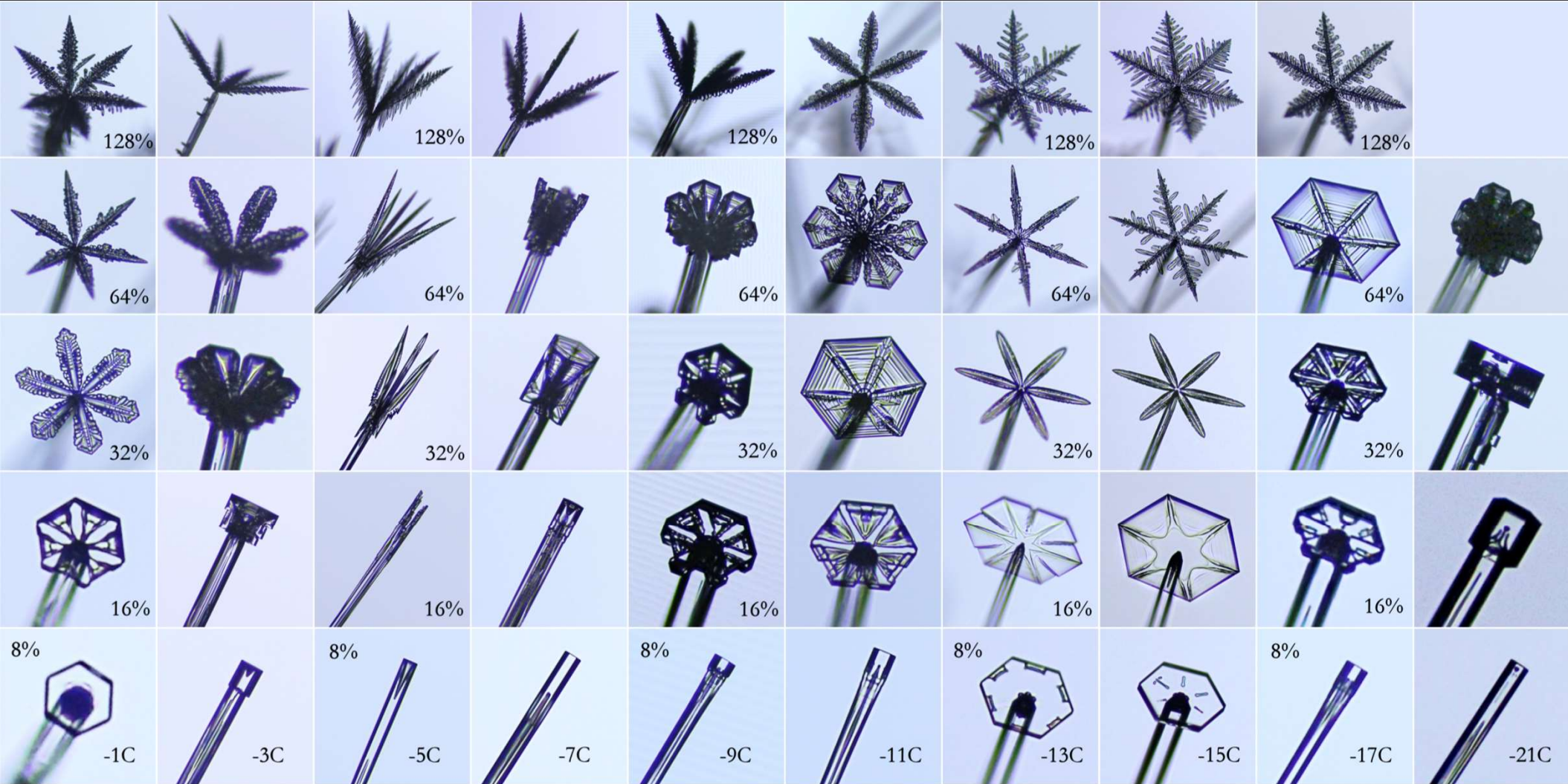
“ELECTRIC” ICE NEEDLES AS A RESEARCH TOOL



Quantitative 2D cylindrically symmetric models

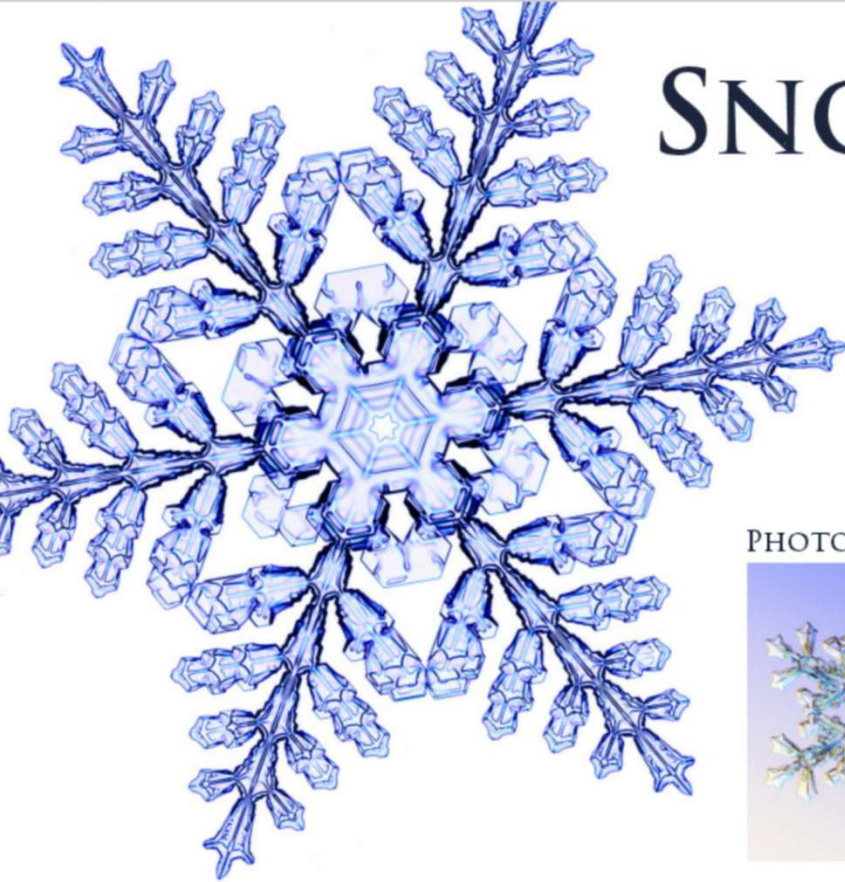


Snow crystals grown on ice needles – Diffusion-limited growth explains vertical axis, order \rightarrow chaos



All single crystals; all grown at constant conditions

What is underlying physics? Possible to make numerical models?



SNOW CRYSTALS

.COM

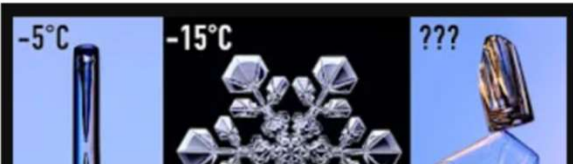
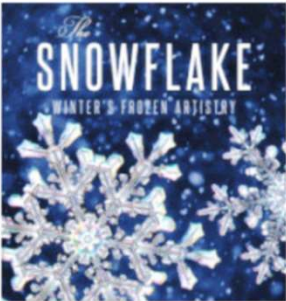
"How full of the creative genius is the air
I should hardly admire more if real stars
--H

Welcome to SnowCrystals.com!
Your online guide to snowflakes, snow crystals, a

PHOTOS



BOOKS



<< Check out this entertaining YouTube video about snow-crystal science, made by the good folks at **Veritasium**.

>> You can find the FULL story about the science of snow crystal formation in my magnum opus at right (weighting in at 456 pages)

